

Helping contaminants emerge: The application of high-resolution mass spectrometry to non-targeted analysis of organic pollutants

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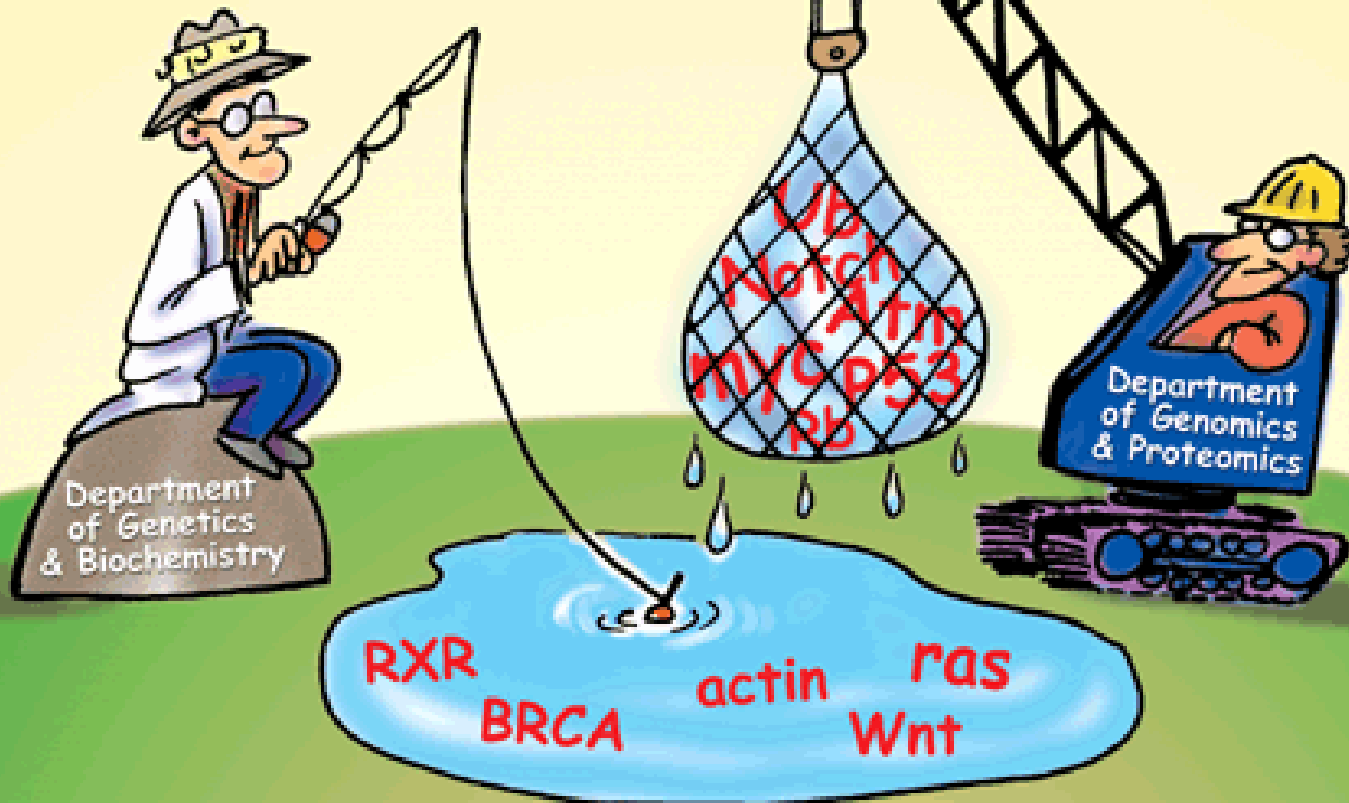
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forging a sustainable future

What are the next emerging contaminants and how can we find them in the environment?

Environmental
Analytical Chemist:
1970s - 2010

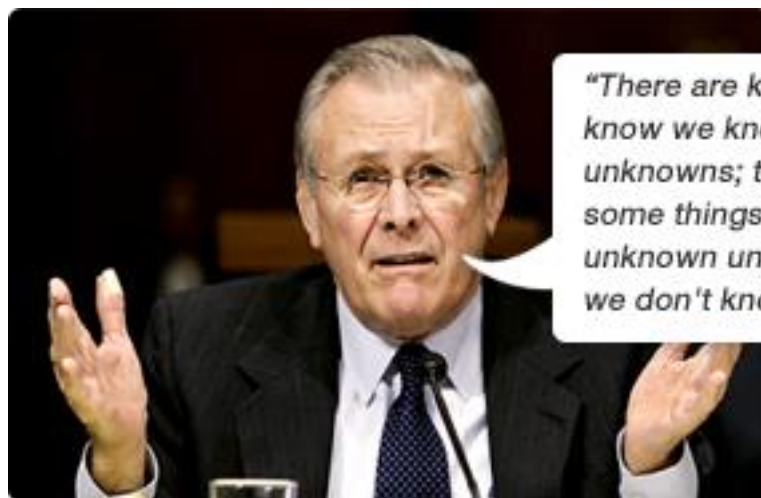
Environmental
Analytical Chemist:
2010 & beyond



Science 16 February 2001: vol. 291 no. 5507 1221-1224

LC-HRMS: An emerging technique for “helping contaminants emerge”

LC-MS strategies for characterization of organic contaminants			
Screening technique:	<i>Targeted</i>	<i>Suspect</i>	<i>Non-target</i>
Question:	<i>Are compounds x, y, & z present in this sample?</i>	<i>Which compounds of a defined list are present in this sample?</i>	<i>Which compounds are present in this sample?</i>
Compound Types:	<i>Known-knowns</i>	<i>Known-unknowns</i>	<i>Known-unknowns & unknown-unknowns</i>



“There are known knowns; there are things we know we know. We also know there are known unknowns; that is to say we know there are some things we do not know. But there are also unknown unknowns – the ones we don’t know we don’t know.”

Wastewater is a significant source of emerging contaminants to the aquatic environment

- Micropollutant fate in wastewater treatment is process-dependent
- Removal efficiencies may vary substantially depending on micropollutant structure
- Needed: methods for “holistic” assessment of micropollutant fate during wastewater treatment

OBJECTIVE:

Application of a non-targeted LC-HRMS method for fate-dependent analysis of micropollutants in wastewater and surface water

Study site and sampling

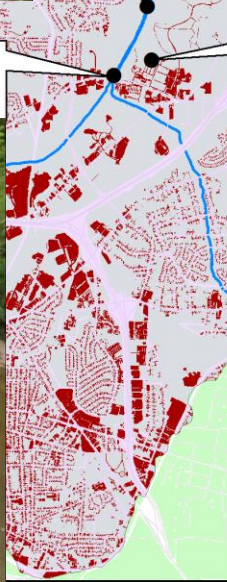
Ellerbe Creek downstream of North Durham WWTP



Ellerbe Creek upstream of North Durham WWTP

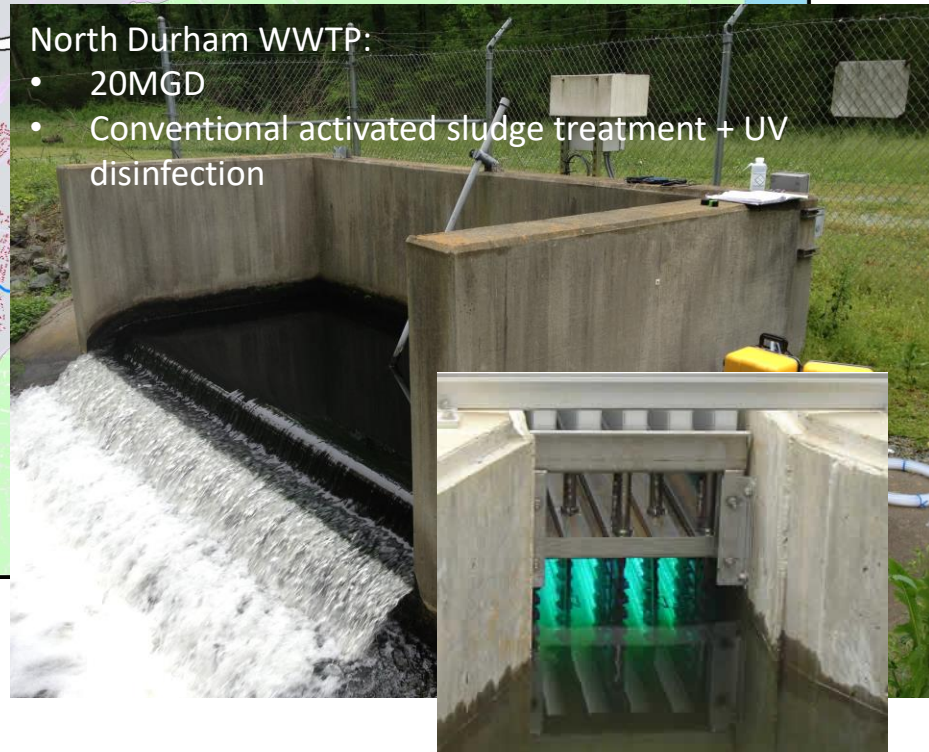


A



North Durham WWTP:

- 20MGD
- Conventional activated sludge treatment + UV disinfection



Sample Preparation and Instrumental Analysis

Sampling:

- Daily grab samples (Tue-Fri)
- Triplicate sampling on one day



Sample enrichment:

- 500 mL sample (primary effluent diluted 1:5 (v/v))
- Spiked with stable isotope labeled standards (19)
- Automated SPE, 500 mg Oasis HLB



UHPLC

- Dionex Ultimate 3000, 100x2.1 Thermo Hypersil Gold aQ
- H₂O/ACN gradient, 95% to 1% H₂O in 55 min, 0.5 mL min⁻¹

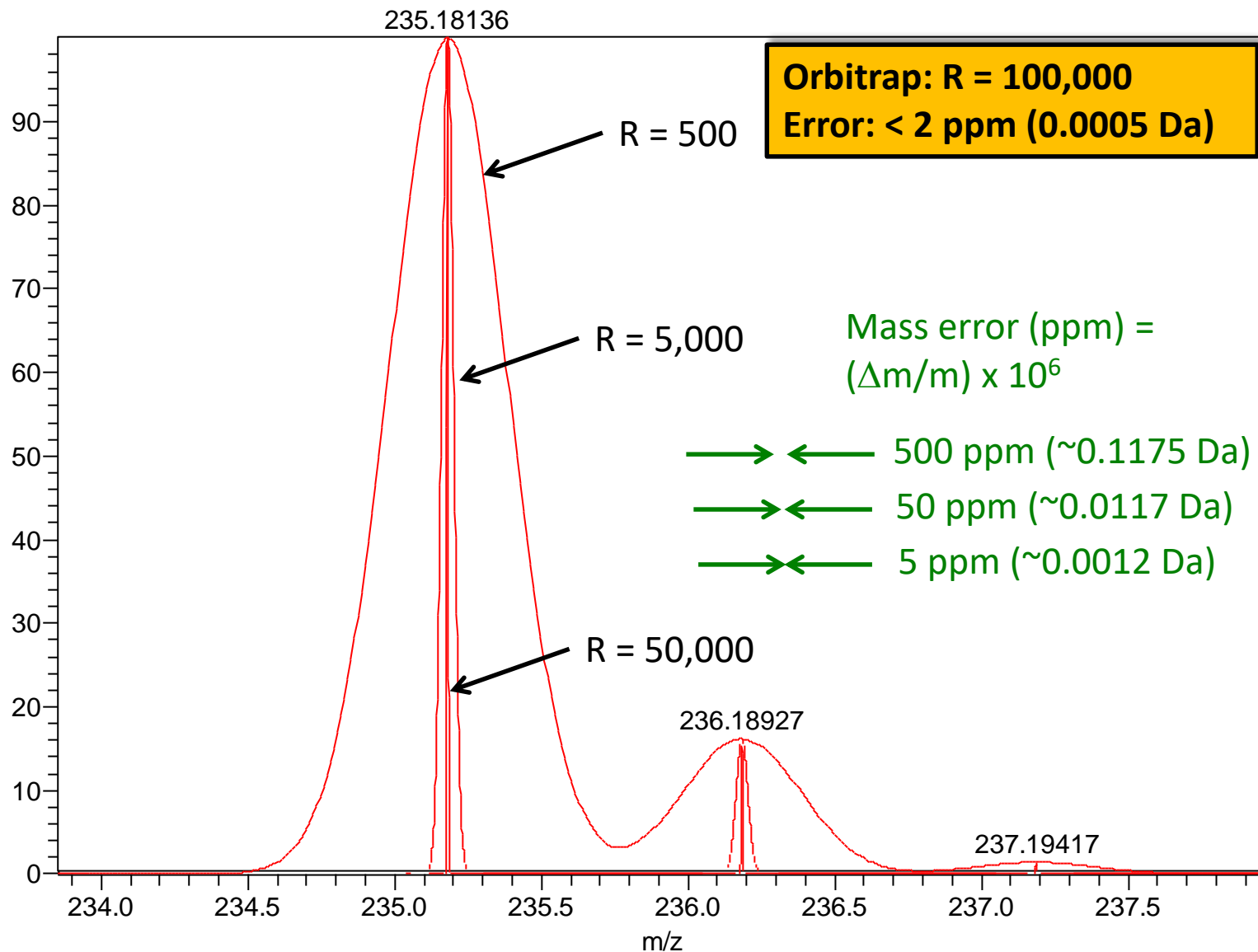


High-resolution mass spectrometry

- Thermo LTQ Orbitrap Velos, ESI(+)
- Full-scan (m/z 100-2000), accurate mass, R=60k FWHM
- Top-4 data-dependent accurate mass MS², R=7500 FWHM



Why do we use HRMS for non-targeted analysis of pollutants?

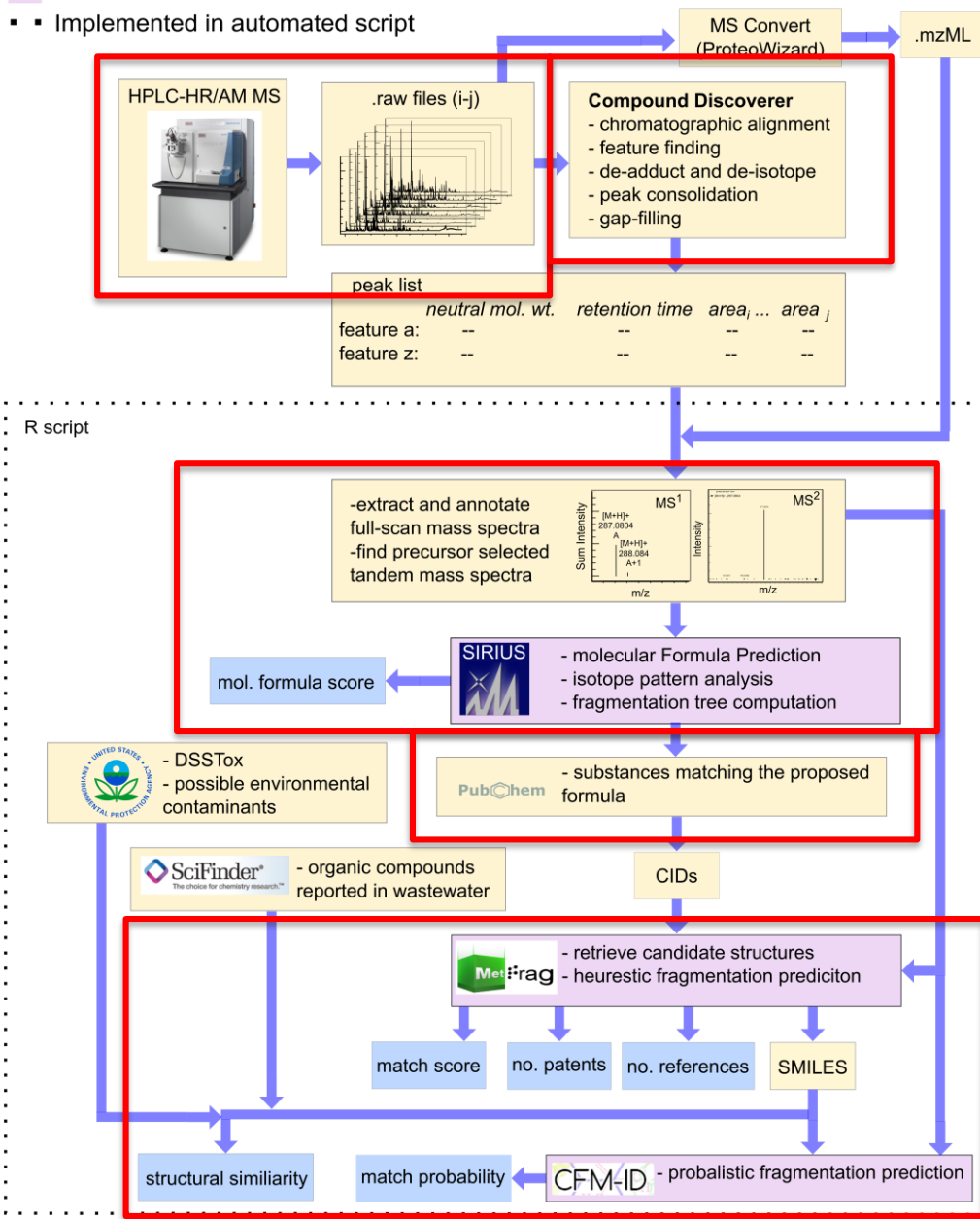


Data analysis workflow

■ Scoring metric

■ Command line tool

■ ■ Implemented in automated script



1. Molecular feature detection

- De-adduct, de-isotope
- Feature = accurate neutral mass + retention time

2. Molecular formula prediction

- Monoisotopic mass decomposition
- Isotope pattern filtering
- Fragmentation tree annotation

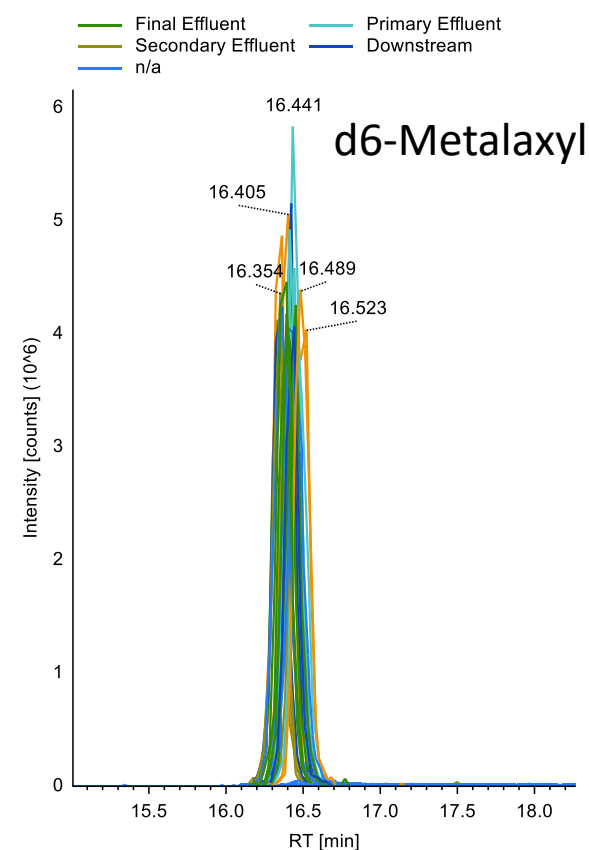
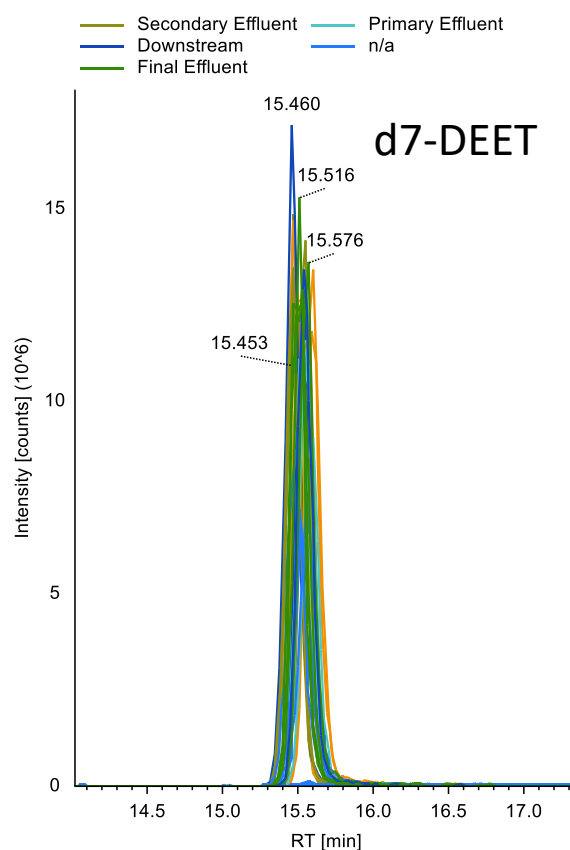
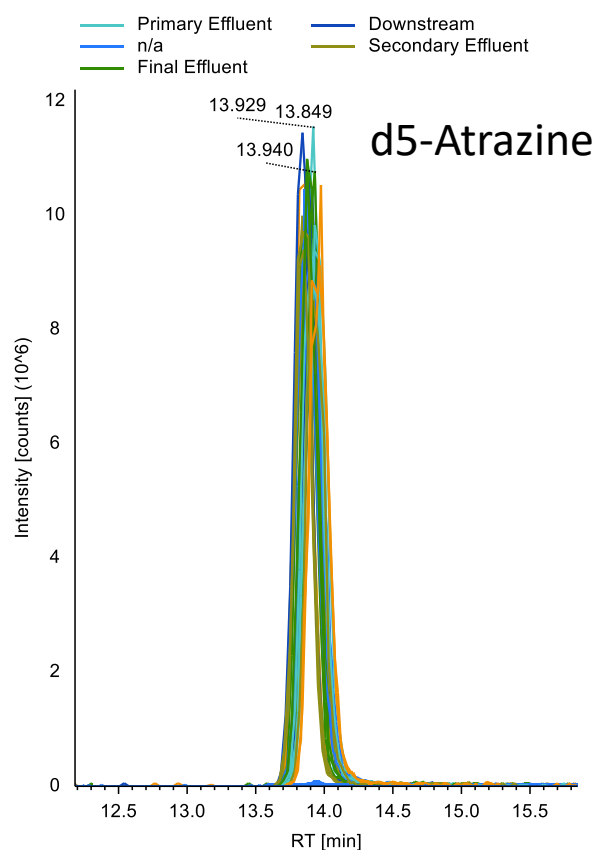
3. Postulate structure

- PubChem formula query

4. Holistic structure scoring

- Combinatorial fragment generator
- Literature and patent data
- Similarity searching

Isotope labeled standards reveal high reproducibility and minimal matrix effects across sample types

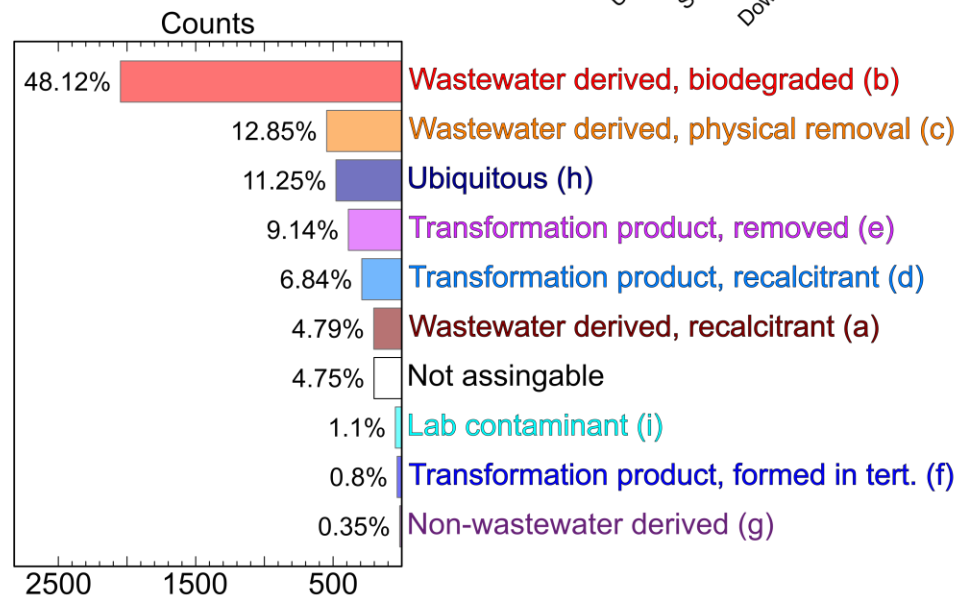
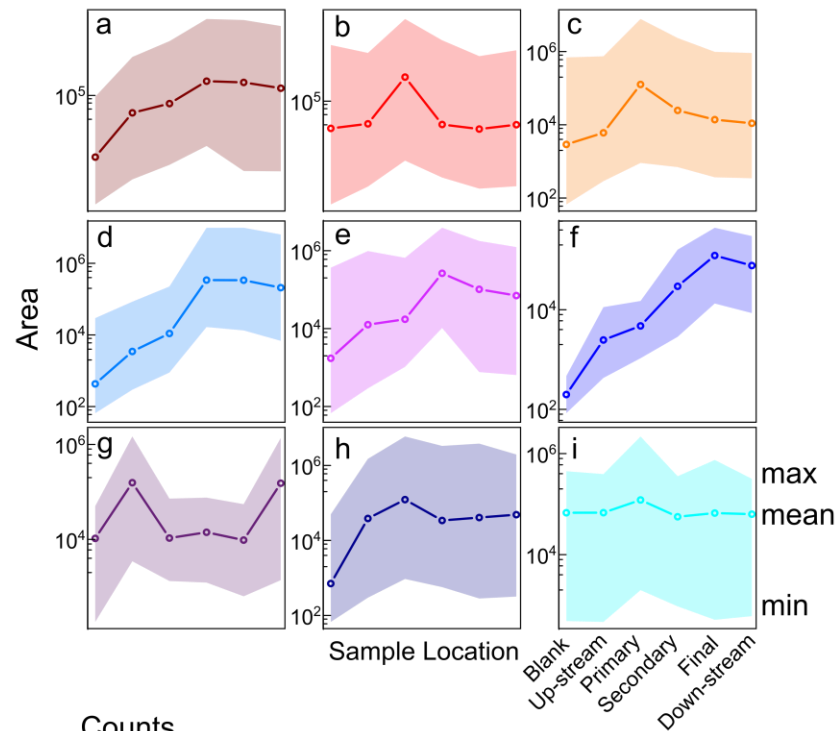
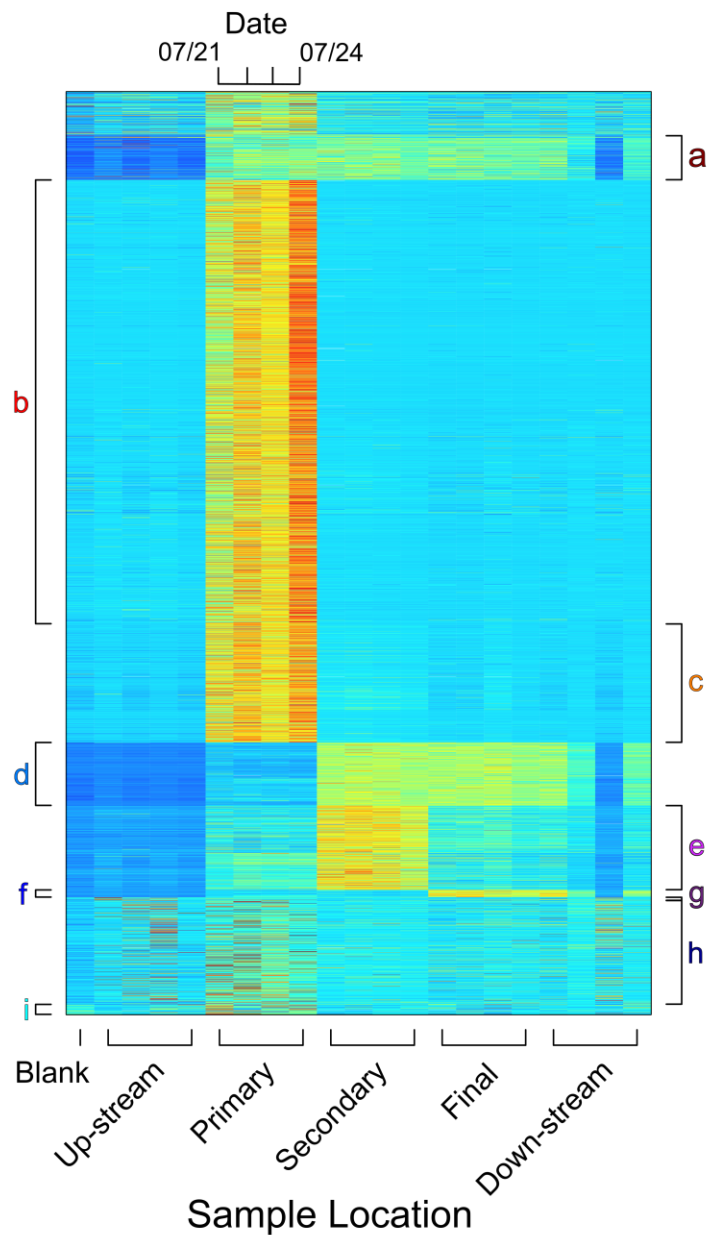


Differential analysis of non-target compounds in wastewater to reveal emerging contaminants

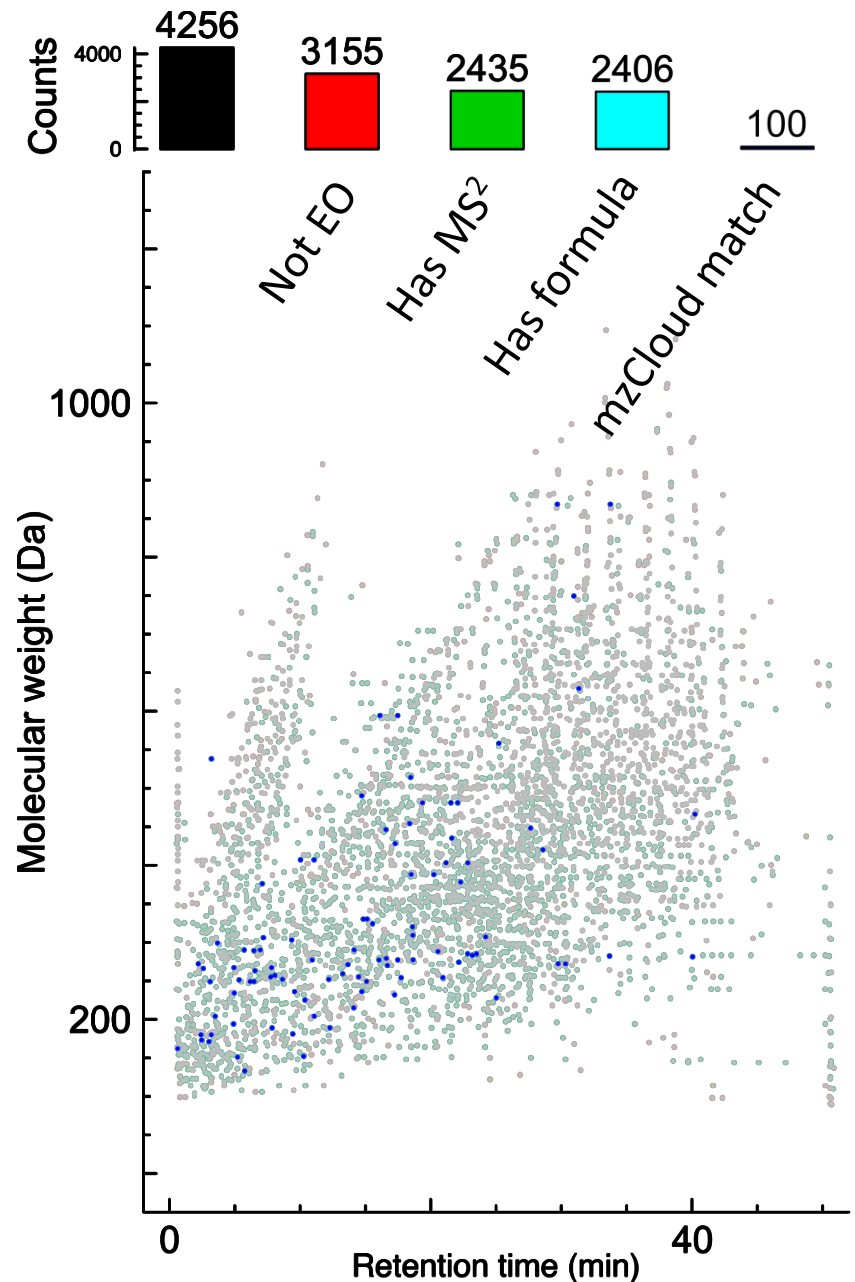
- Screening results were used to generate target lists for differential screening
- ID in screening = mass error $< \pm 2.5$ ppm and $\Delta RT < 0.5$ min

Category	Blank	Upstream	Primary eff.	Secondary eff.	Final eff.	Downstream
Wastewater-derived, biodegradable	○	○	↑	↓	→	→
Wastewater-derived, physical removal	○	○	↑	↘	↘	→
Wastewater-derived, recalcitrant	○	○	↑	→	→	→
Transformation product, removed	○	○	○	↑	↓	→
Transformation product, recalcitrant	○	○	○	↑	→	→
Transformation product, produced in tertiary treatment	○	○	○	↗	↑	→
Non wastewater-derived	○	↑	○	○	○	↑
Ubiquitous	○	→	→	→	→	→
Laboratory contaminant	→	→	→	→	→	→

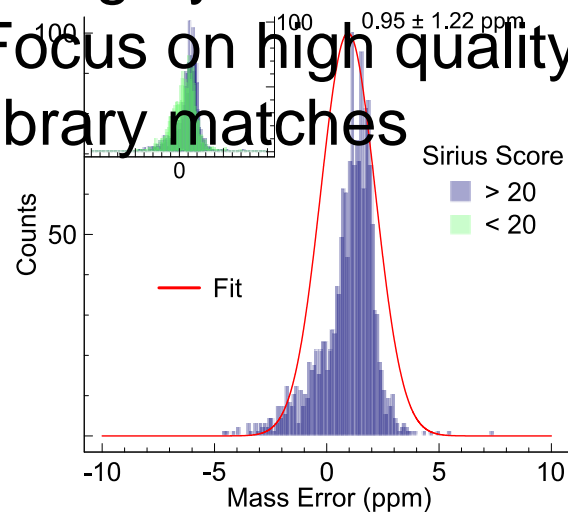
Fate-dependent feature prioritization



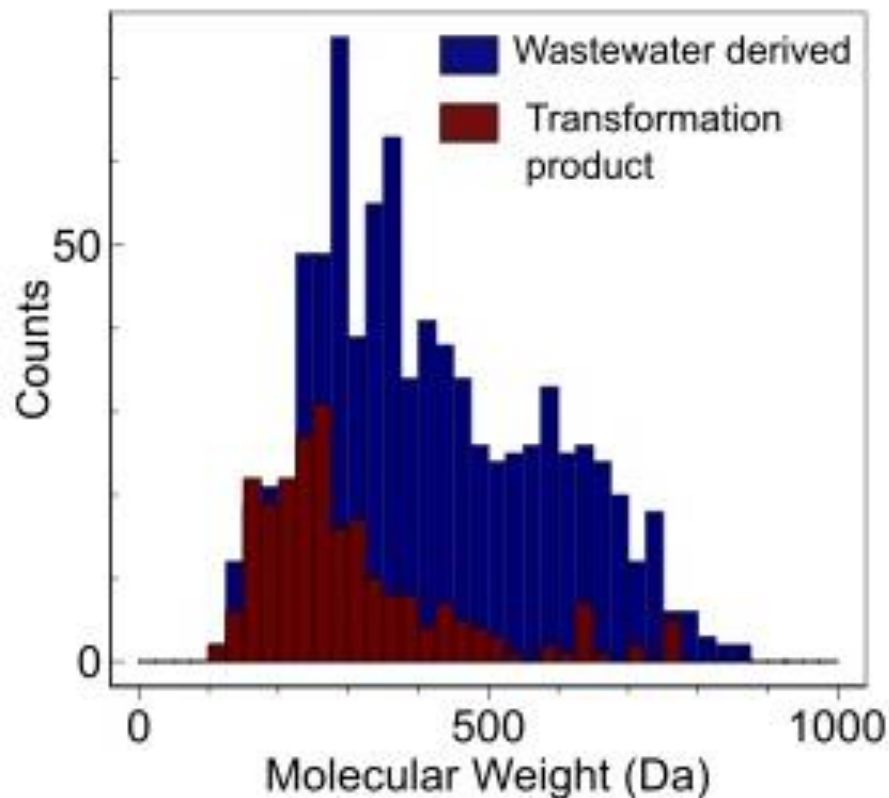
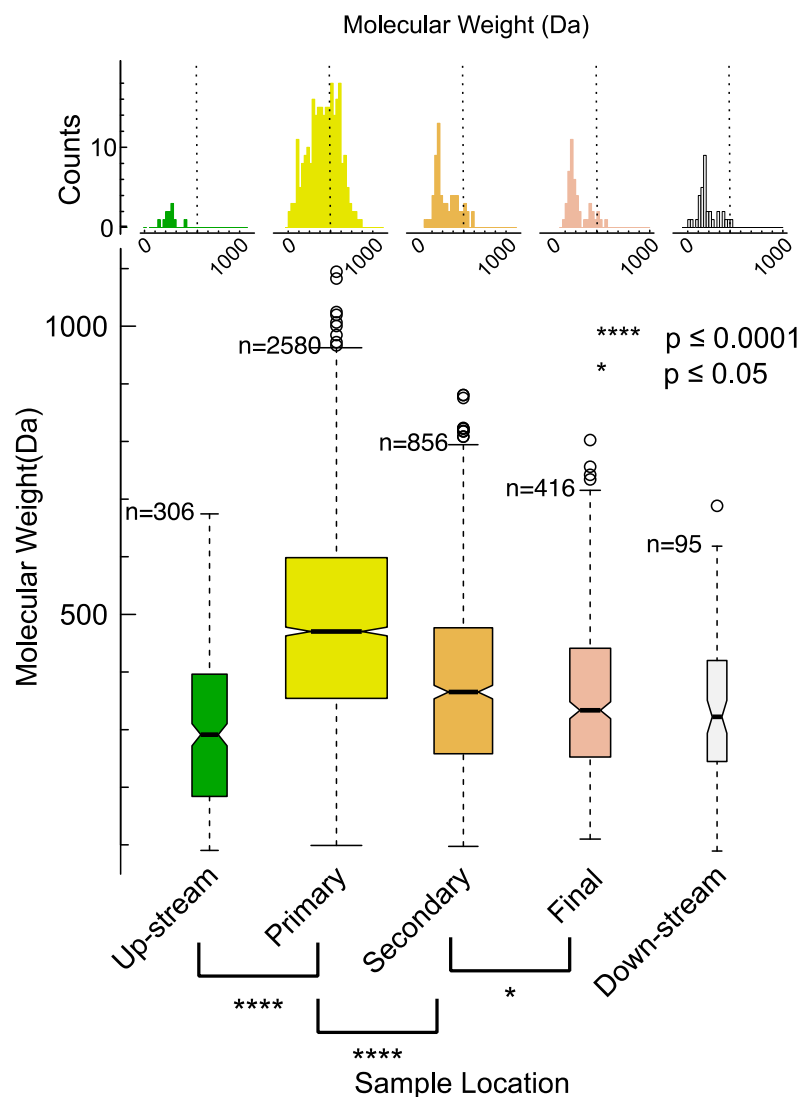
Data filtering



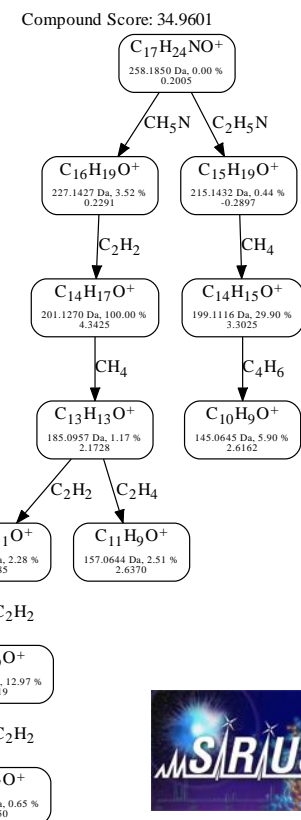
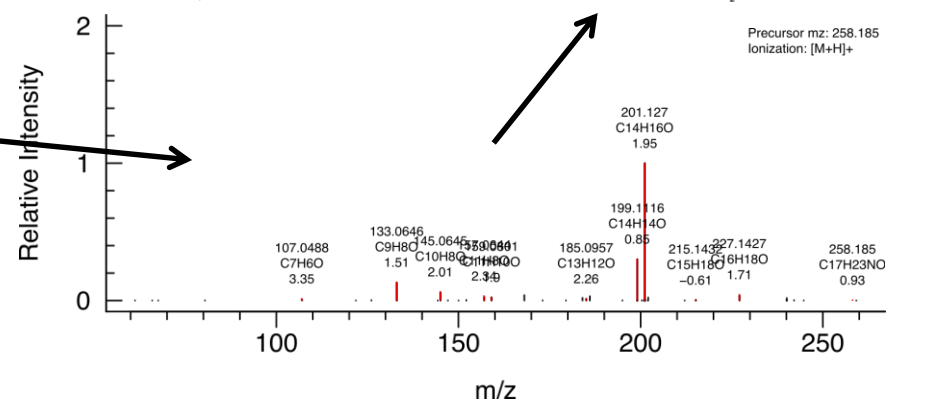
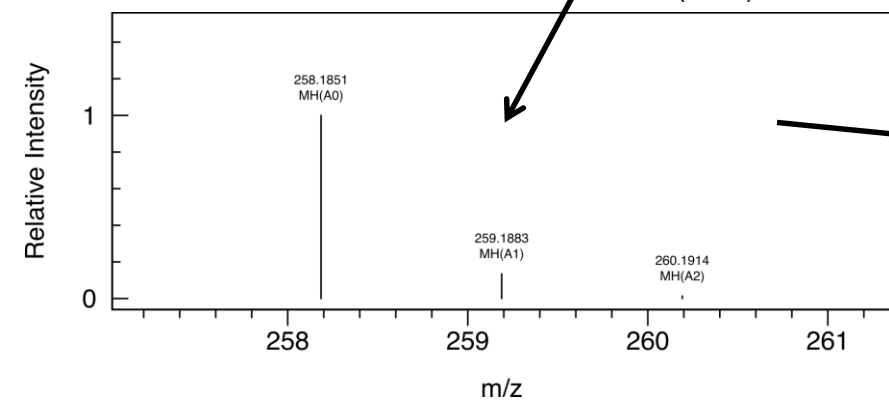
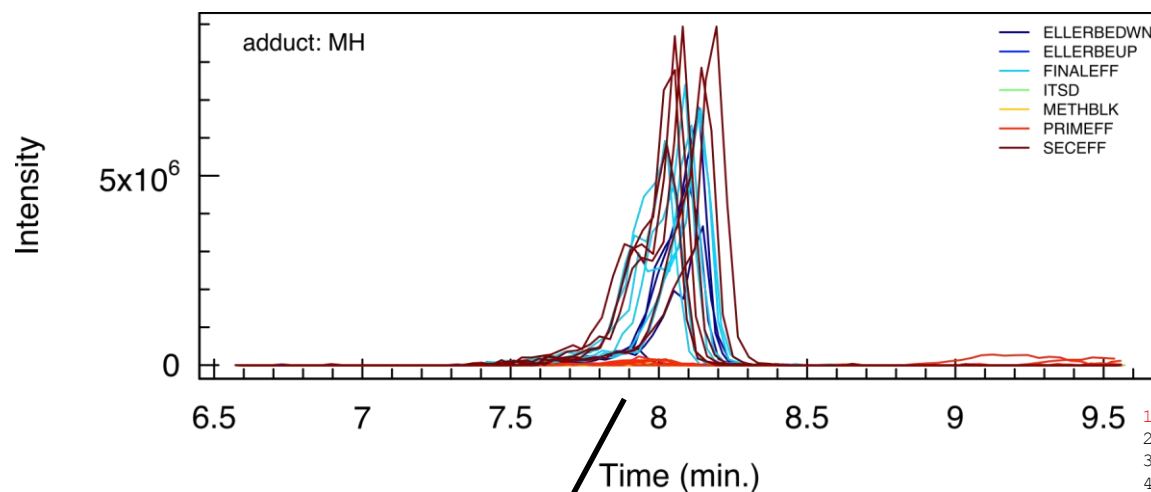
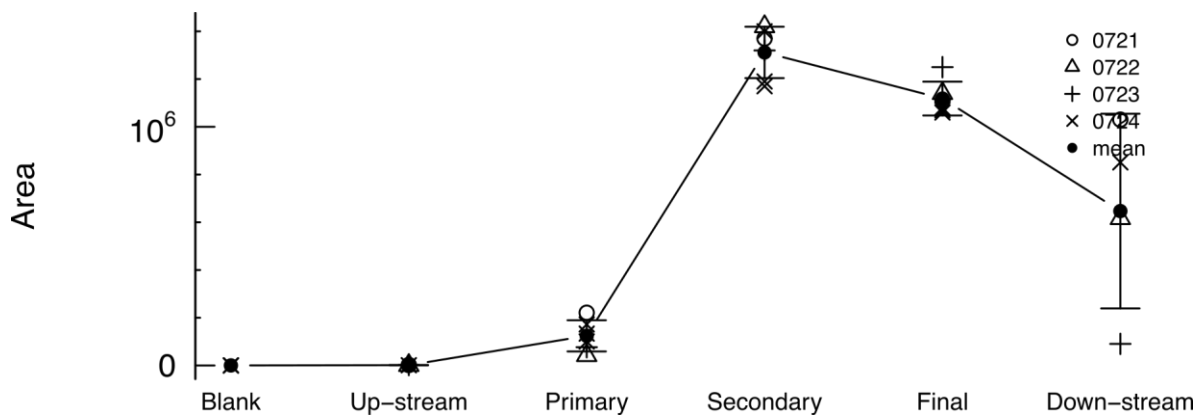
- Remove features corresponding to ethoxylated oligomers
- Filter based on MS² spectra
- Retain features with molecular formula assignment
- Prioritize based on differential category
- Focus on high quality mzCloud library matches



Results: Feature molecular weight distributions change during the wastewater treatment process



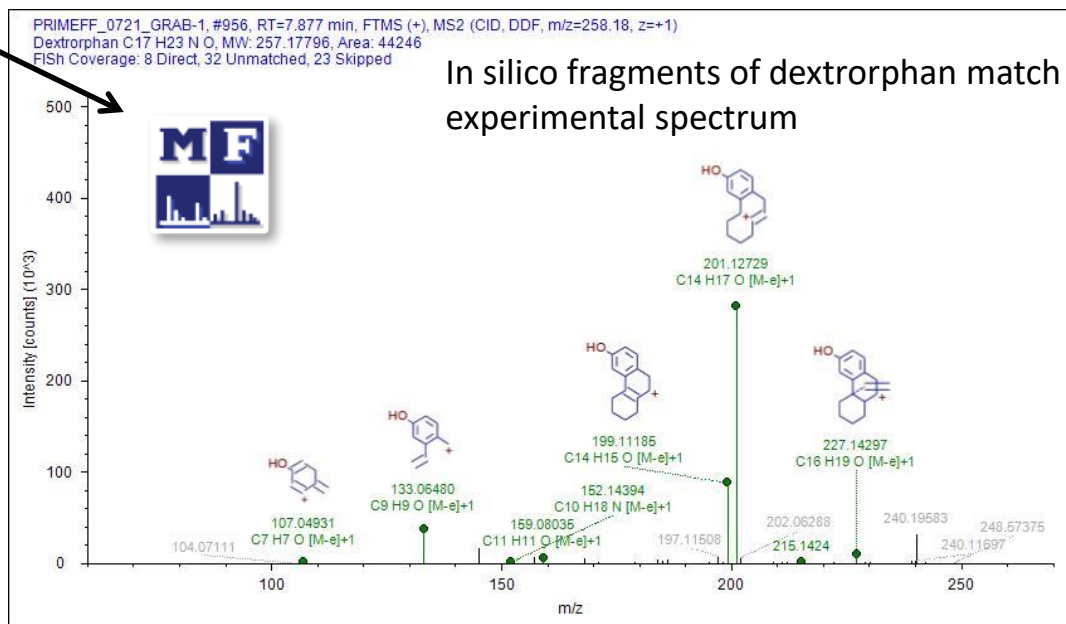
257.1777 Da feature follows a recalcitrant transformation product profile



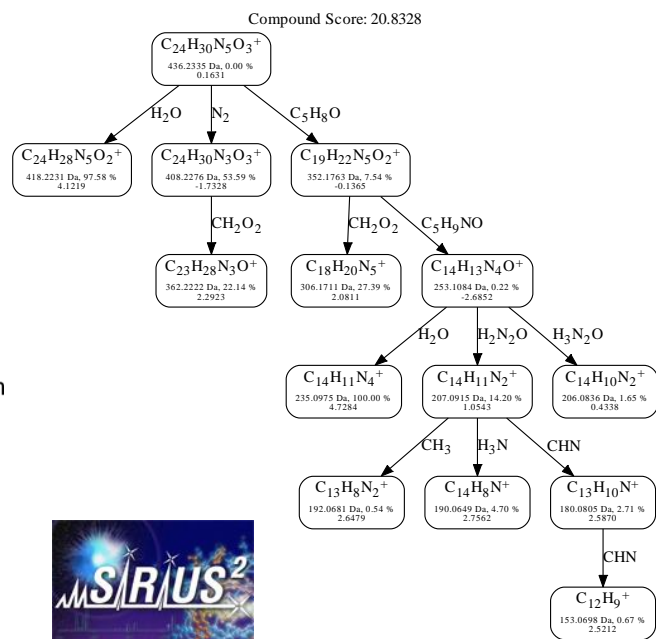
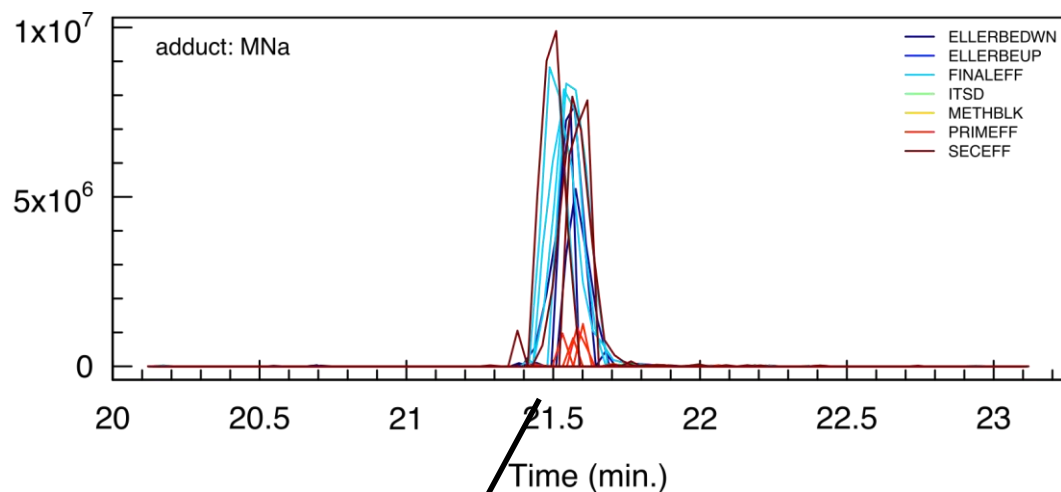
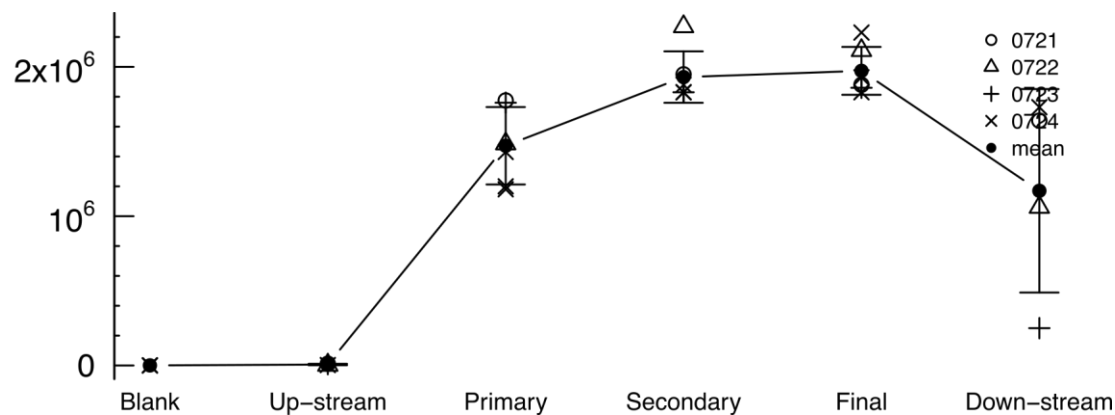
- 1.) C17H23NO score: 34.96 tree: +28.10 iso: 6.86 peaks: 11 91.33 %
- 2.) C13H26N2OP score: 23.35 tree: +15.03 iso: 8.33 peaks: 13 93.83 %
- 3.) C12H22FN4O score: 21.51 tree: +12.90 iso: 8.61 peaks: 11 88.13 %
- 4.) C7H21F2N7O score: 6.13 tree: +6.13 iso: 0.00 peaks: 11 91.26 %

Dextrorphan is a known, activated human metabolite of the cough suppressant dextromethorphan

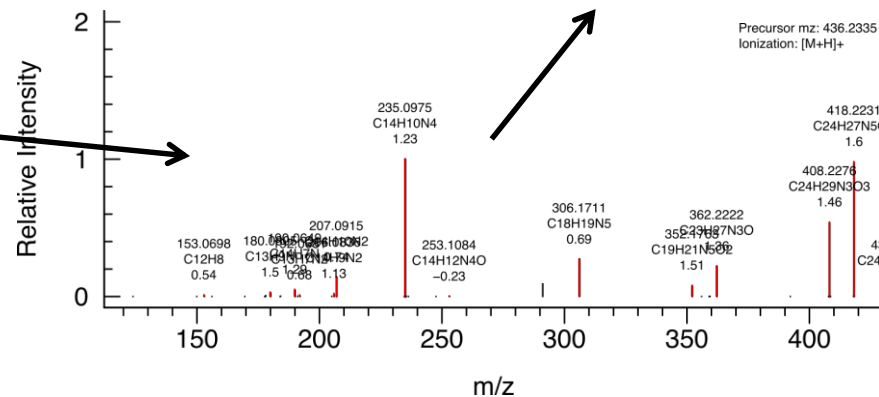
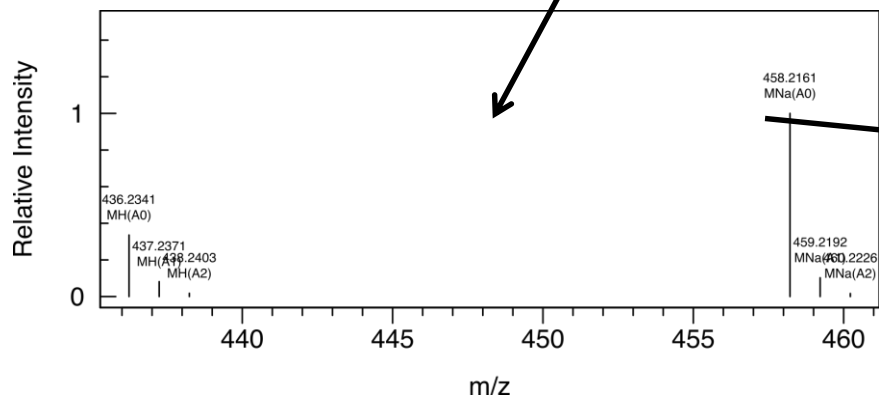
CSID	Mass	References	Structure	Name
16736212	102	40		Dextromethorphan



A feature at 435.2244 Da is wastewater-derived and recalcitrant

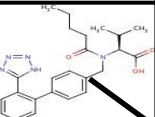
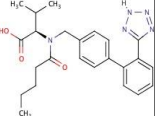
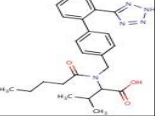
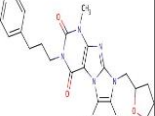
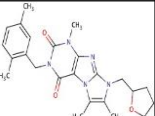


- 1.) $C_{24}H_{29}N_5O_3$ score: 20.83 tree: +20.83 iso: 0.00 peaks: 14 95.67 %
- 2.) $C_{22}H_{27}N_8O_3$ score: 15.72 tree: +15.72 iso: 0.00 peaks: 15 95.87 %
- 3.) $C_{19}H_{28}FN_8O_3$ score: 13.65 tree: +13.65 iso: 0.00 peaks: 14 95.67 %
- 4.) $C_{16}H_{33}N_7O_5S$ score: 13.08 tree: +13.08 iso: 0.00 peaks: 12 85.63 %
- 5.) $C_{19}H_{29}F_4N_5O_2$ score: 12.88 tree: +12.88 iso: 0.00 peaks: 13 74.11 %
- 6.) $C_{18}H_{30}N_9O_2P$ score: 12.26 tree: +12.26 iso: 0.00 peaks: 11 64.07 %
- 7.) $C_{21}H_{31}F_4N_2O_3$ score: 12.00 tree: +12.00 iso: 0.00 peaks: 14 95.40 %
- 8.) $C_{19}H_{30}ClN_9O$ score: 11.25 tree: +11.25 iso: 0.00 peaks: 10 63.87 %
- 9.) $C_{15}H_{33}F_2N_4O_8$ score: 10.95 tree: +10.95 iso: 0.00 peaks: 14 95.67 %
- 10.) $C_{15}H_{31}FN_9O_3P$ score: 9.57 tree: +9.57 iso: 0.00 peaks: 13 80.27 %



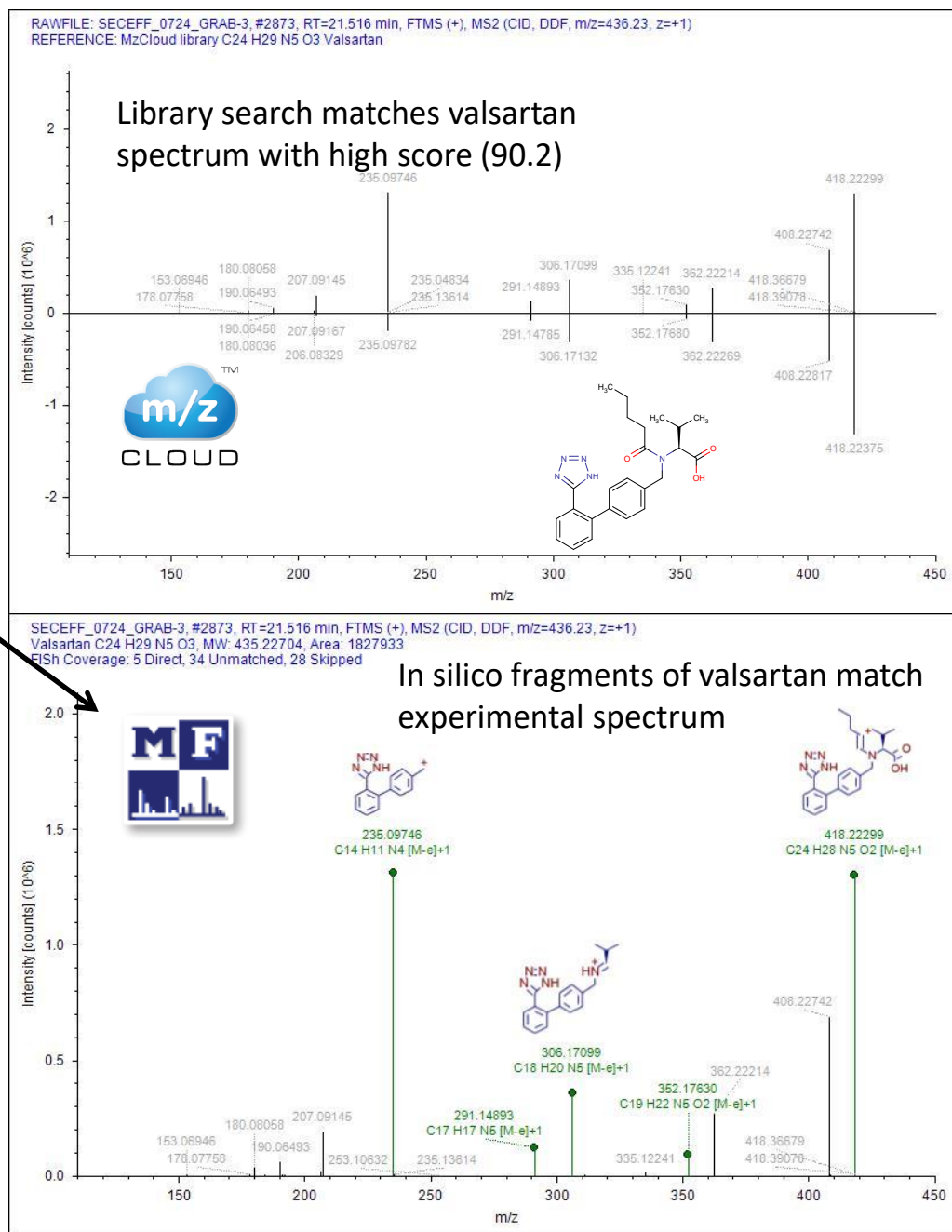
Valsartan tentatively identified by library match and *in silico* fragments

Chemspider matches for $C_{24}H_{29}N_5O_3$

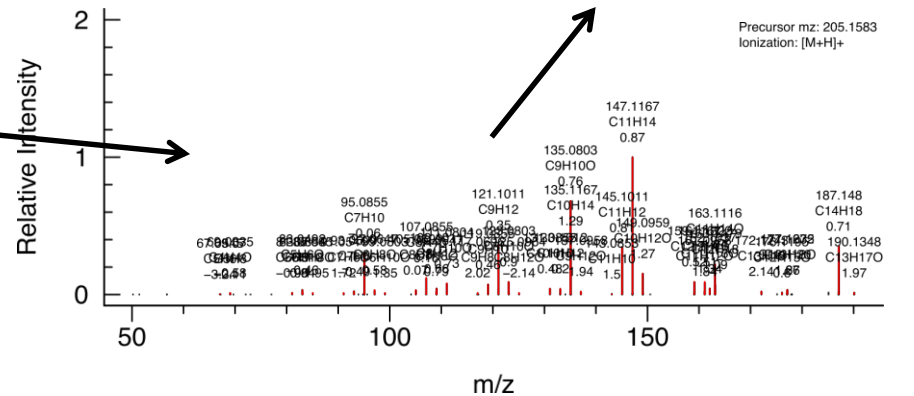
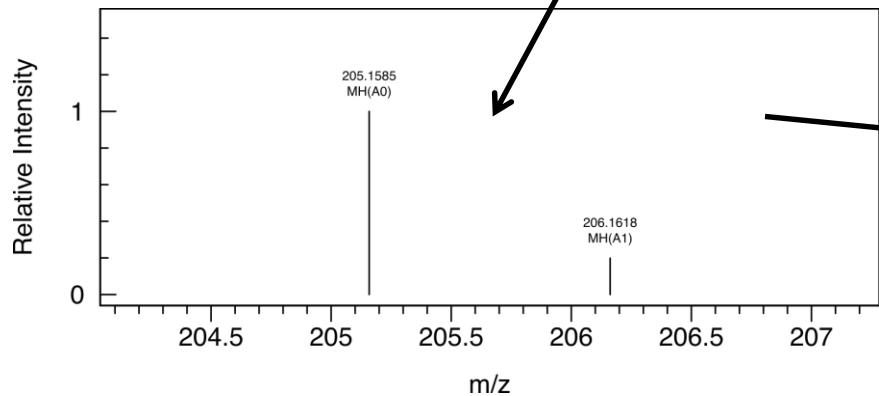
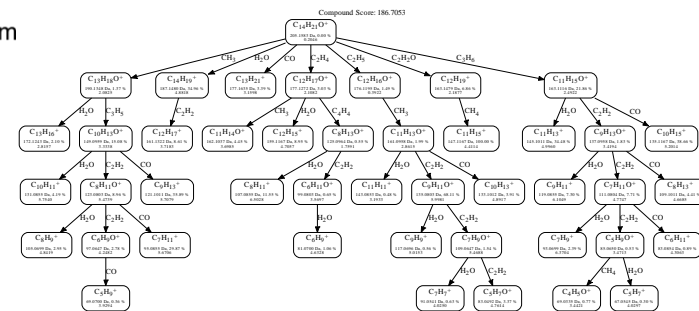
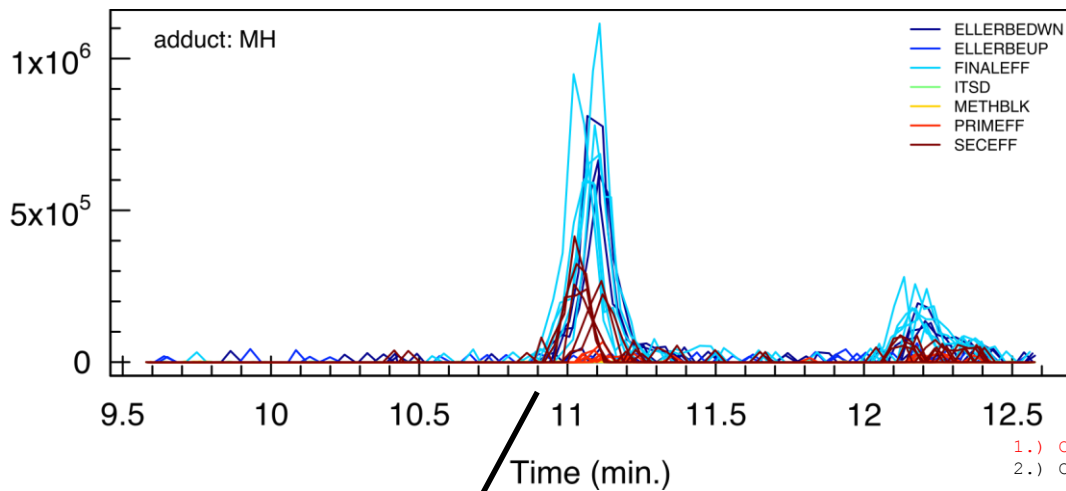
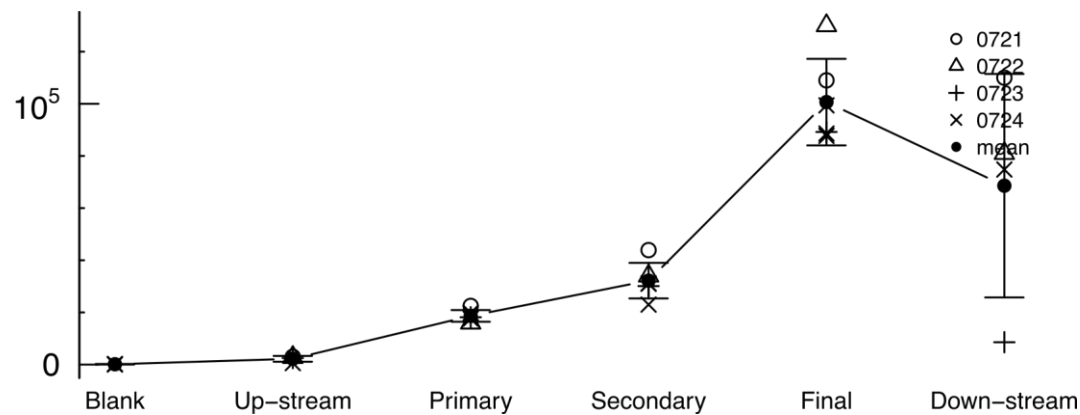
CSID	Δ Mass [ppm]	# References	Structure	Name
54833	1.4	1050		Valsartan
4447678	1.4	34		N-pentanoyl-N-([2'-(1H-tetrazol-5-yl)biphenyl-4-yl]methyl)-D-valine
5448	1.4	26		N-Pentanoyl-N-([2'-(1H-tetrazol-5-yl)-4-biphenyl]methyl)valine
2414607	1.4	15		1,6,7-Trimethyl-3-(3-phenylpropyl)-8-(tetrahydro-2-furanylmethyl)-1H-imidazo[2,1-f]purine-2,4(3H,8H)-dione
2414601	1.4	13		3-(2,5-Dimethylbenzyl)-1,6,7-trimethyl-8-(tetrahydro-2-furanylmethyl)-1H-imidazo[2,1-f]purine-2,4(3H,8H)-dione

Valsartan is a highly prescribed angiotensin II receptor antagonist and has previously been shown to be recalcitrant to biodegradation in wastewater treatment.

Bergheim, M. et al. 2014. *Environ. Chem.*, 11, 431-444



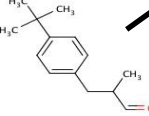
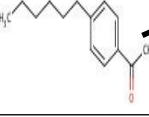
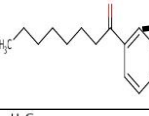
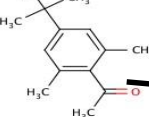
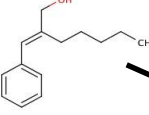
204.1512 Da feature increases after tertiary treatment (UV)



1.) C14H20O score: 186.71 tree: +179.28 iso: 7.42 peaks: 44 98.87 %
2.) C9H19FN3O score: 154.34 tree: +154.34 iso: 0.00 peaks: 44 98.87 %

Without library matches, *in silico* fragmentation provides structural clues

Chemspider matches for $C_{14}H_{20}O$

CSID	Δ Mass [ppm]	# References	Structure	Name
199342	1.14	99		Protectol PP (Lilial)
110058	1.14	67		p-Hexylacetophenone
66895	1.14	52		Octanophenone
67442	1.14	32		1-(4-tert-Butyl-2,6-dimethylphenyl)ethanone
4519770	1.14	27		(2E)-2-Benzylidene-1-heptanol

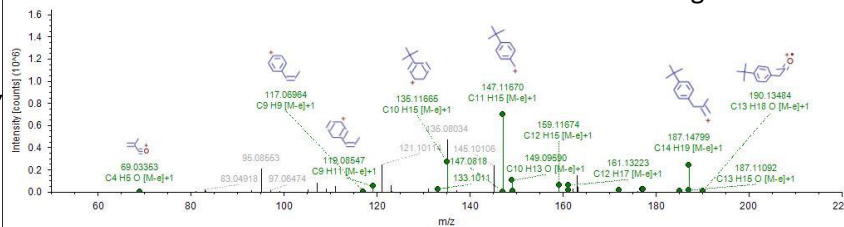


C8 oxo-alkylbenzenes have been reported as oxidation products of petroleum hydrocarbons in marine environments.

Harvey, G.R. 1995. *Mar. Poll. Bull.*, 30(6), 425-426

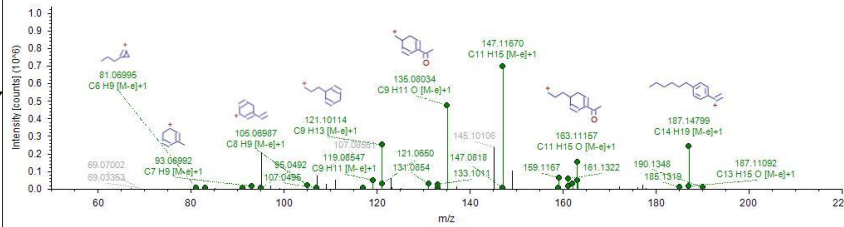
FINALEFF_0724_GRAB-3, #1389, RT=11.054 min, FTMS (+), MS2 (CID, DDF, m/z=205.16, z=+1)
Benzeneacetaldehyde C14 H20 O, MW: 204.15142, Area: 88567
FISH Coverage: 19 Direct, 52 Unmatched, 22 Skipped

FISH coverage = 26.76



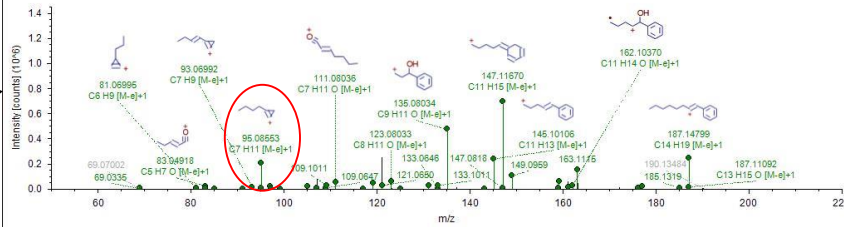
FINALEFF_0724_GRAB-3, #1389, RT=11.054 min, FTMS (+), MS2 (CID, DDF, m/z=205.16, z=+1)
p-Hexylacetophenone C14 H20 O, MW: 204.15142, Area: 88567
FISH Coverage: 28 Direct, 43 Unmatched, 22 Skipped

FISH coverage = 39.44



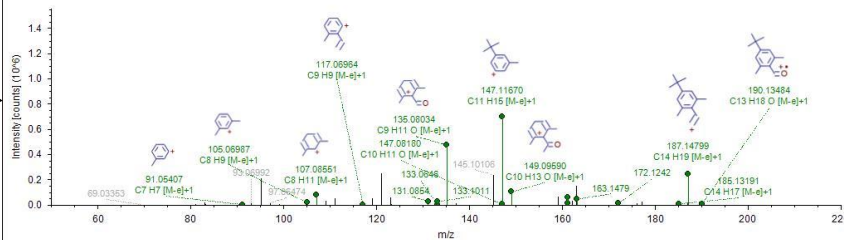
FINALEFF_0724_GRAB-3, #1389, RT=11.054 min, FTMS (+), MS2 (CID, DDF, m/z=205.16, z=+1)
Octanophenone C14 H20 O, MW: 204.15142, Area: 88567
FISH Coverage: 41 Direct, 30 Unmatched, 22 Skipped

FISH coverage = 57.75



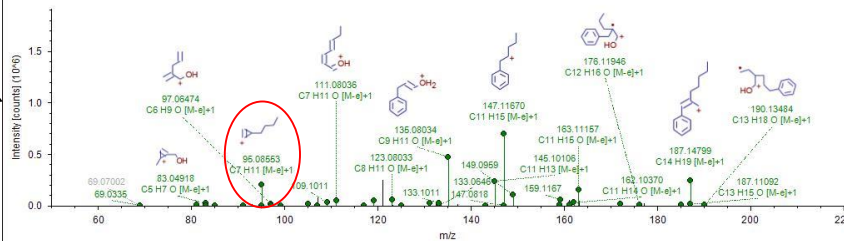
FINALEFF_0724_GRAB-3, #1389, RT=11.054 min, FTMS (+), MS2 (CID, DDF, m/z=205.16, z=+1)
1-(4-tert-Butyl-2,6-dimethylphenyl)ethanone C14 H20 O, MW: 204.15142, Area: 88567
FISH Coverage: 18 Direct, 53 Unmatched, 22 Skipped

FISH coverage = 25.35

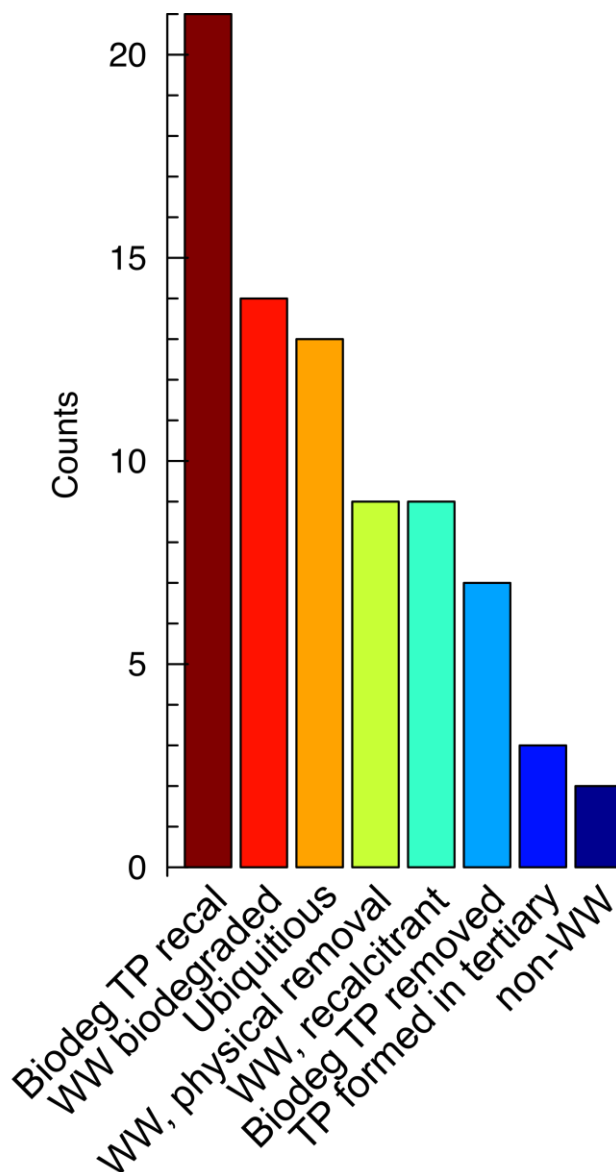
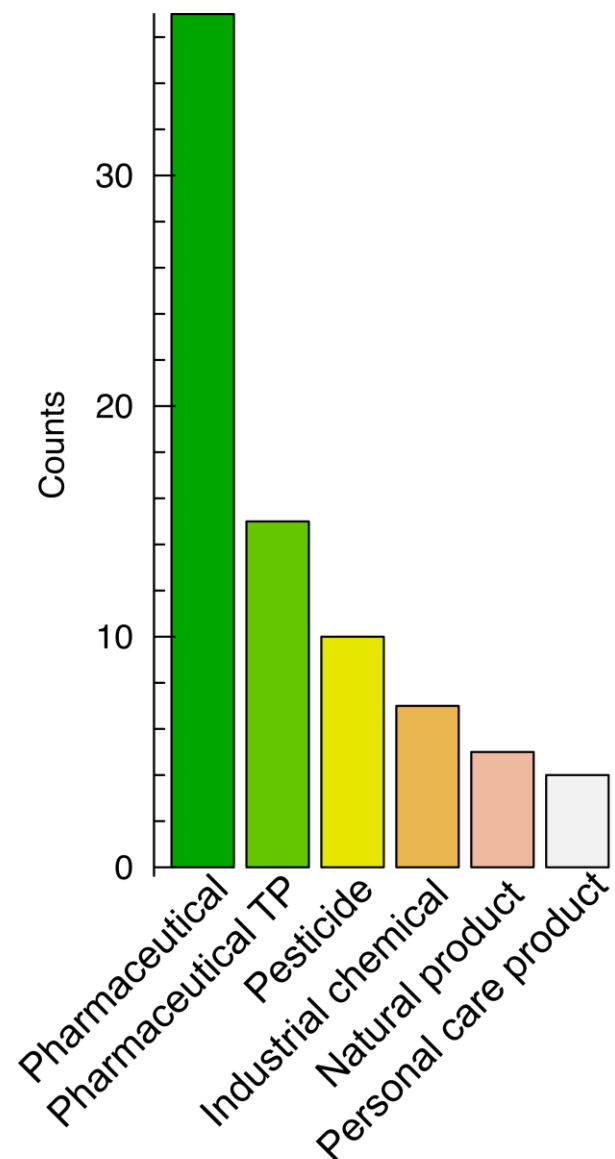


FINALEFF_0724_GRAB-3, #1389, RT=11.054 min, FTMS (+), MS2 (CID, DDF, m/z=205.16, z=+1)
2-Benzylidene-1-heptanol C14 H20 O, MW: 204.15142, Area: 88567
FISH Coverage: 39 Direct, 32 Unmatched, 22 Skipped

FISH coverage = 54.93



Results: Suspect compounds tentatively identified



- 78 compounds from six classes were tentatively identified in wastewater/surface water
- 20 of these were confirmed with standards (100% correct assignment)
- A further 1,101 features were annotated as polyethoxylated surfactants (comprising 39% of features identified as wastewater-derived, biodegradable)
- The largest fraction of ID'd compounds was classified as recalcitrant transformation products.

Conclusions: Non-targeted analysis of micropollutants in wastewater

- LC-HRMS coupled with optimized non-targeted screening workflows provide essential tools for conducting “fate-directed analysis” of organic contaminants in the environment.
- Differential analysis coupled to suspect screening is a powerful approach for identifying treatment-specific profiles of micropollutants during wastewater treatment
- Holistic analysis reveals that wastewater micropollutant burdens change both qualitatively and quantitatively during treatment
- Opportunities exist to utilize this approach to inform future wastewater treatment process design and optimization.

Semivolatile organic contaminants in the indoor environment: a challenging “exposome”

- Research on SVOCs has focused on occurrence and effects in the ambient environment – there have been few comprehensive studies on human exposure indoors
- SVOCs escape from household products over time and may accumulate in the indoor environment
- They are applied to consumer products to enhance performance or durability – such as:



Phthalates in personal care products



Flame retardants in furniture and electronics



Bisphenol A in waterbottles



Surfactants in cleaning agents



Antioxidants in food packaging

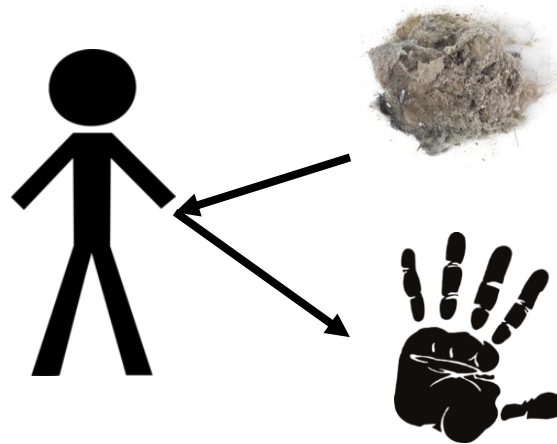
Why study SVOC's indoors?

- Some SVOC's are potential endocrine disruptors
 - Bisphenol A is a xenoestrogen
 - Flame retardants have been shown to act on the thyroid hormone receptor



87% of our time
is spent indoors

Exposure through: inhalation,
ingestion, dermal absorption,

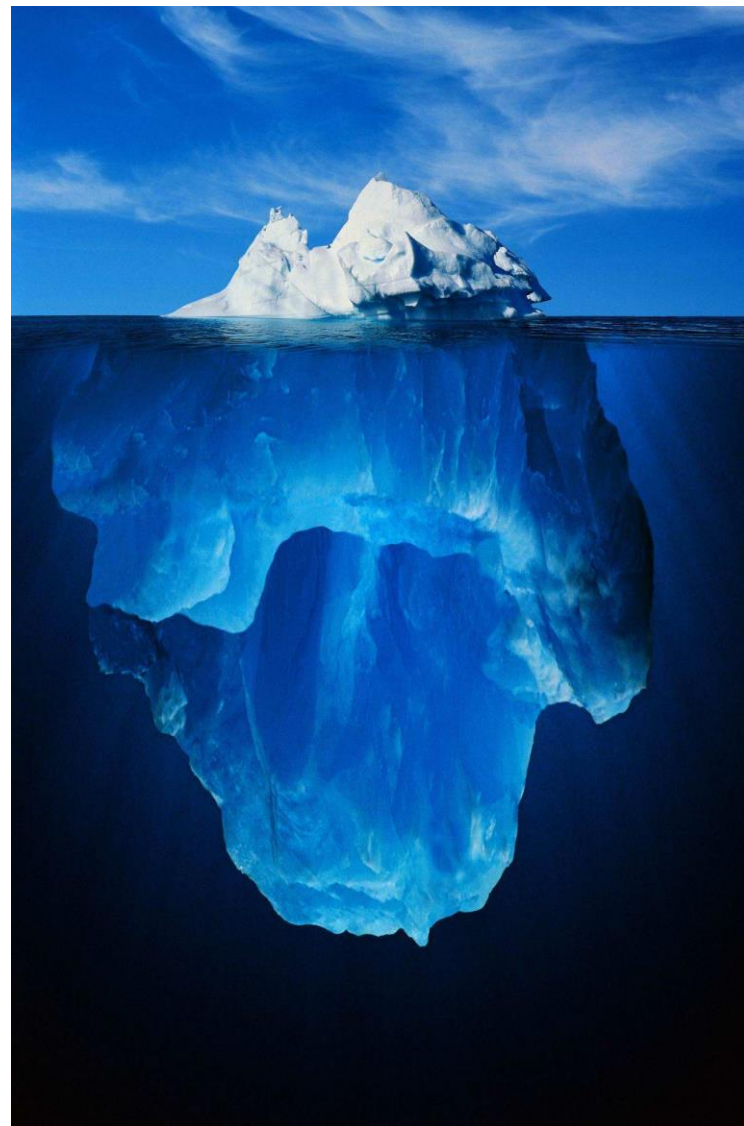


Objective:

Assess human exposure to SVOCs from the indoor environment through non-targeted analysis of paired house dust and hand wipes samples.

Analytical strategy for dust and handwipe samples

- Most indoor exposure analysis has applied gas chromatography mass spectrometry (focus on nonpolar organic contaminants)
- Liquid chromatography coupled with high resolution mass spectrometry can be used to characterize (semi)polar organic contaminants within indoor environments.
- Non-targeted data analytics allows *de novo* identification, prioritized by compounds with highest exposure potential.
- This approach complements more targeted, quantitative analysis of SVOCs by LC-MS/MS or GC-MS approaches.





10 x dust and handwipes
+ dust blanks and wipe blanks

Sample preparation

Extraction by sonication in Hexane/Dichloromethane 1:1;
Solvent exchange to 10 % Acetonitrile in H₂O by speedvac, sonication and centrifugation.

Liquid Chromatography

Reversed phase separation C18,
From 10 % Acetonitrile to 99% in 60 min

Comprehensive 2D Liquid Chromatography

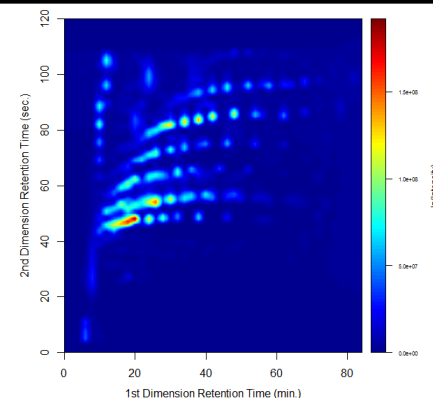
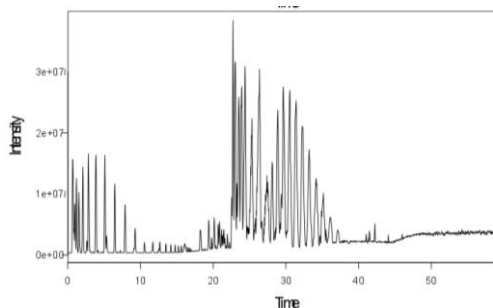
Size exclusion X reversed phase separation
90 min run divided into 2 min segments

Orbitrap Velos

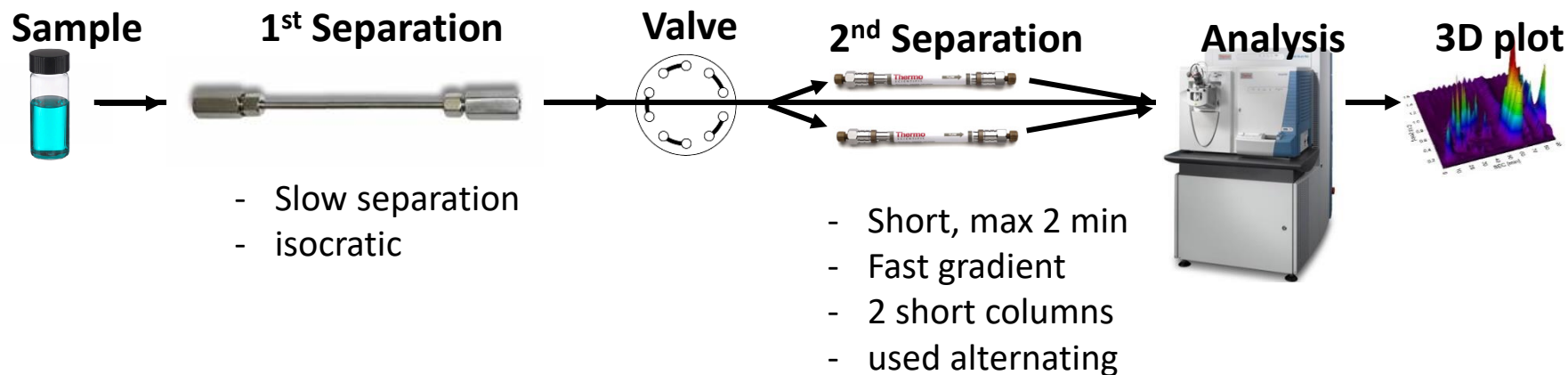
ESI(+) and ESI (-)
Resolution: 60'000 @ m/z 400
Top 4 data dependent MSMS
CID with 35 normalized energy

Orbitrap Velos

ESI(+)
Resolution: 60'000 @ m/z 400



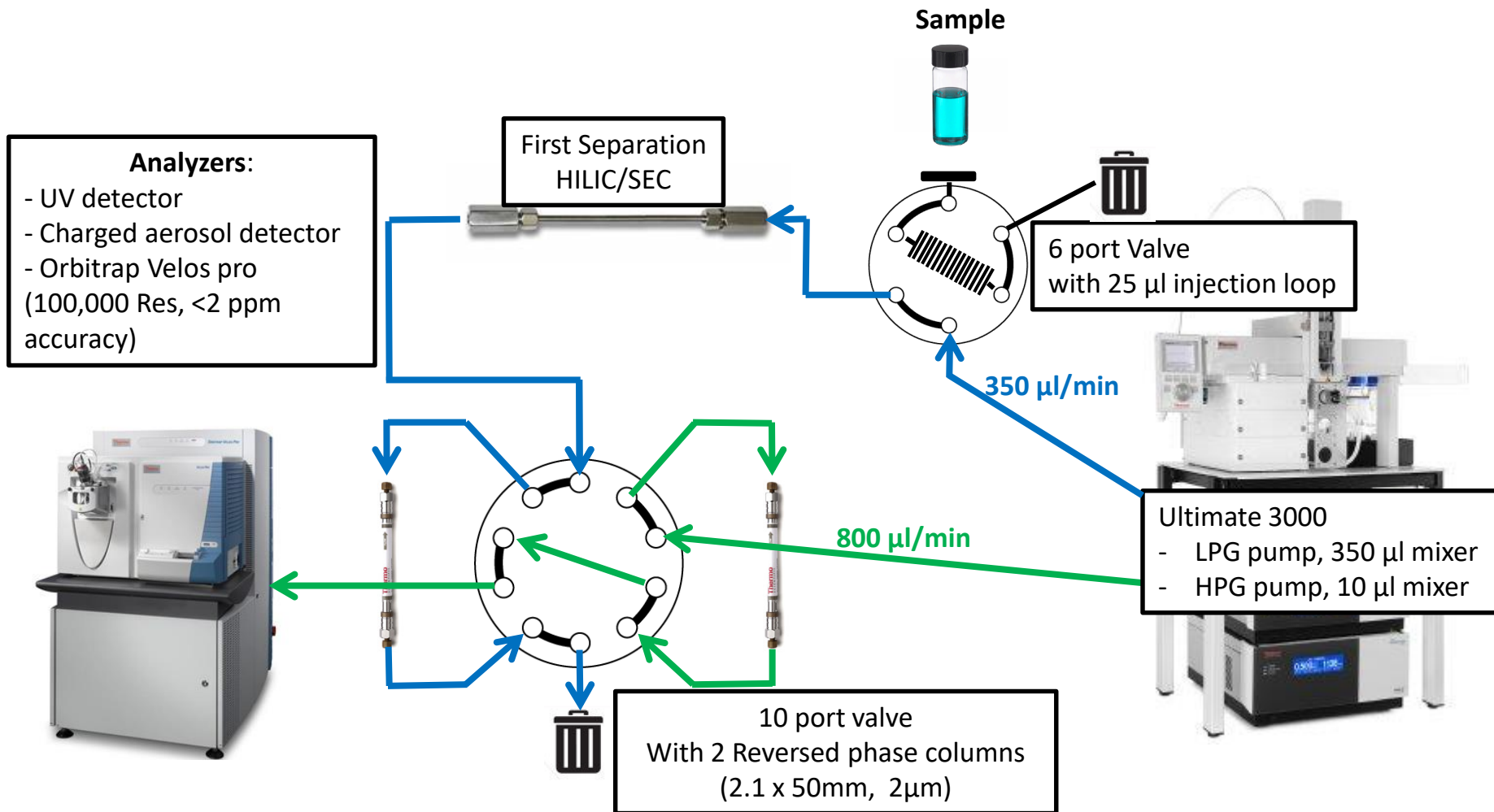
Comprehensive 2D UHPLC (LC x LC)



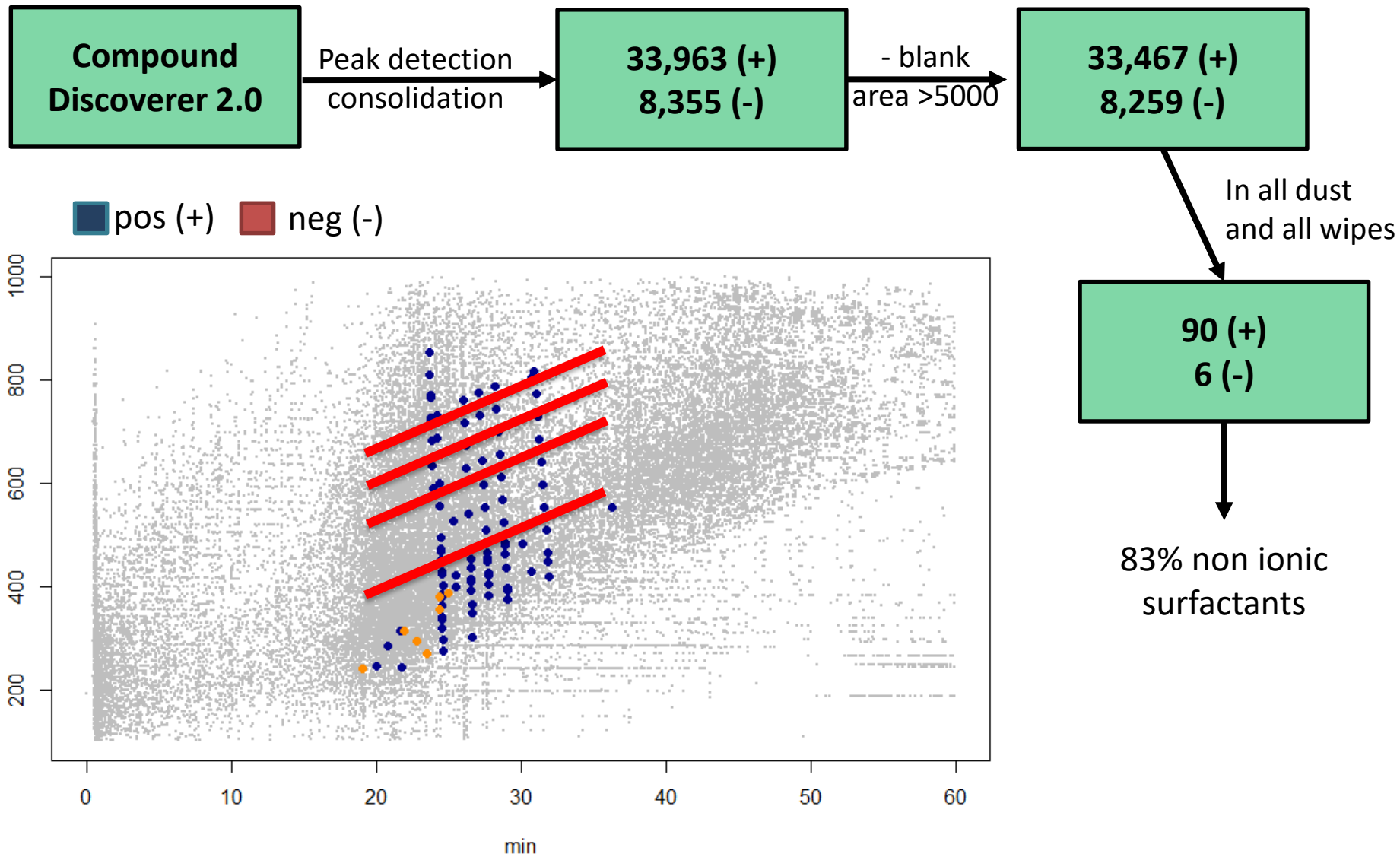
For effective separation:

- Separation mechanisms must be **orthogonal**.
- Example: Size and Hydrophobicity or Hydrophilic interaction and Hydrophobicity.
- While eluting from the first column – requires strong retention on the second column

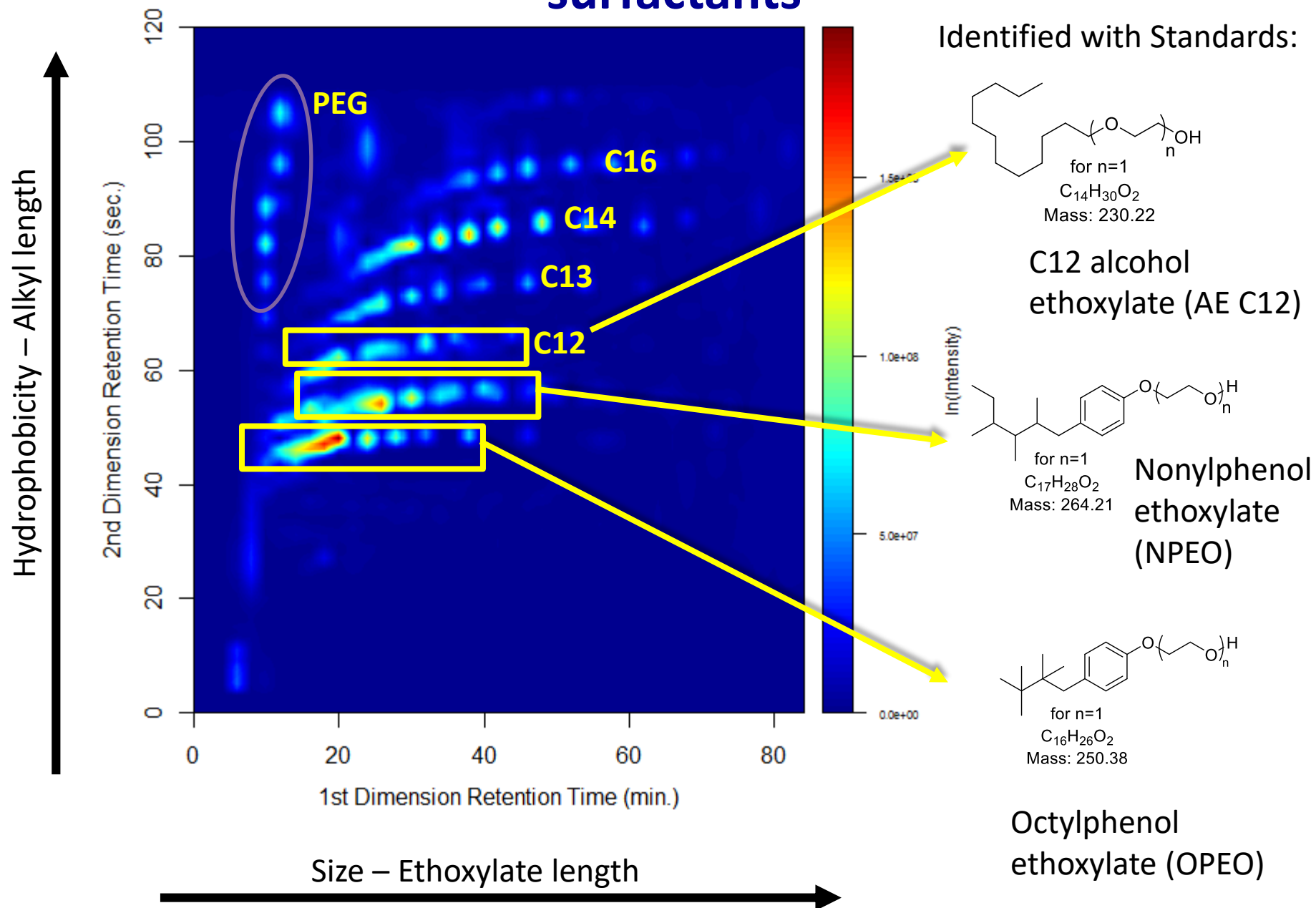
2D UHPLC-HRMS configuration



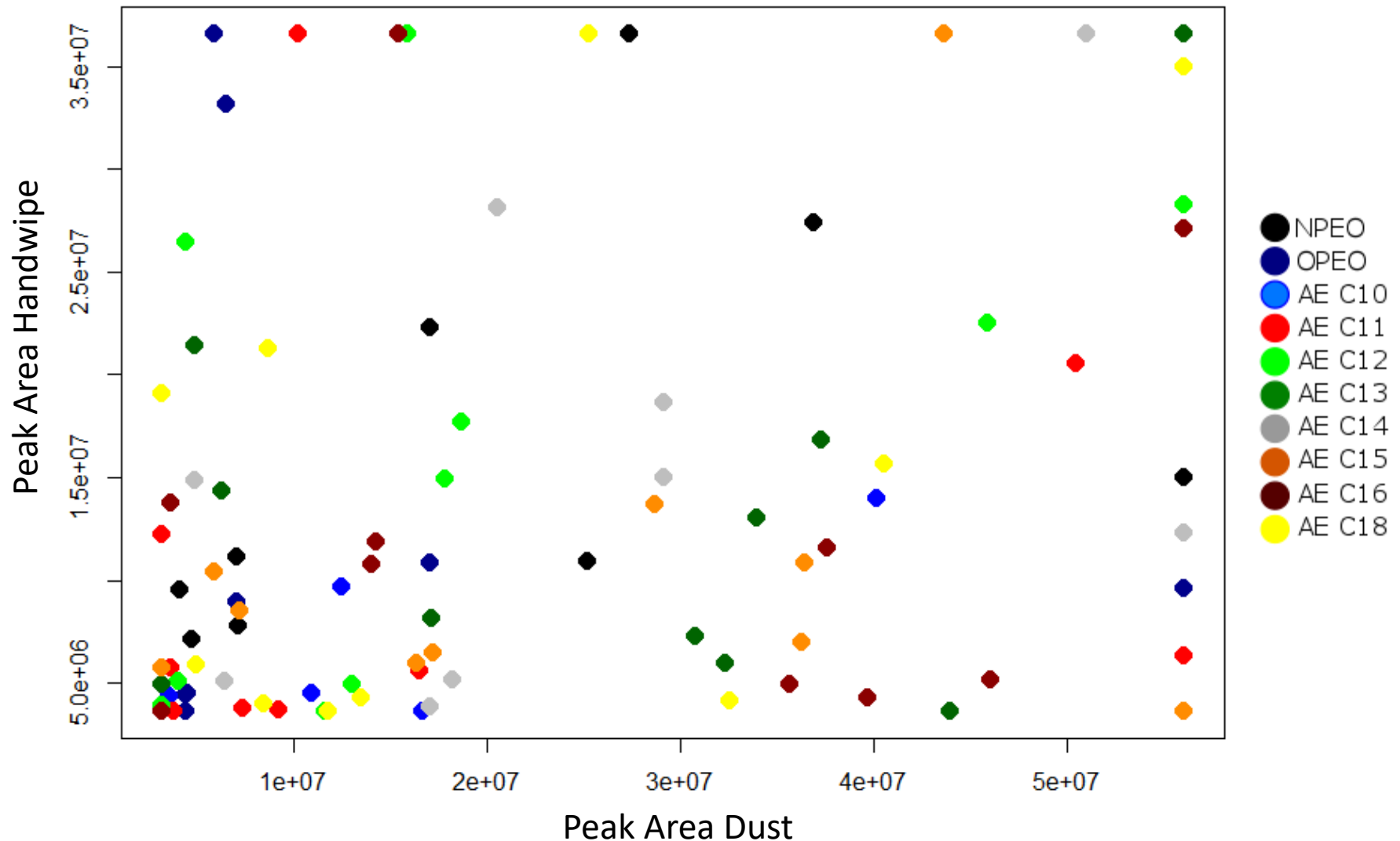
Data processing starts with Thermo Compound Discoverer 2.0 for peak consolidation/filtering



Comprehensive LC x LC-HRMS of dust reveals ethoxylated surfactants

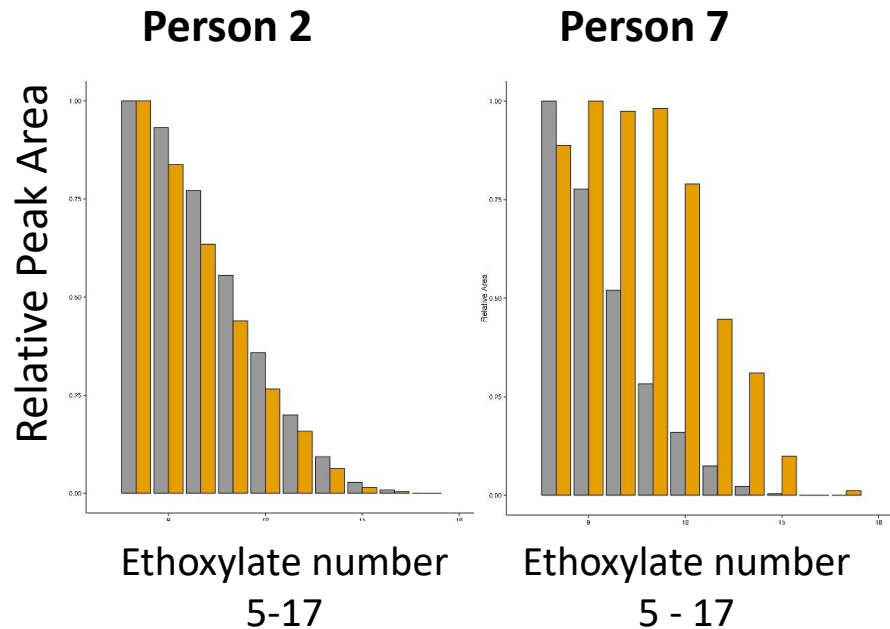


There was no correlation between ethoxylated surfactant peak areas in paired dust/handwipe samples (decoupled sources?)

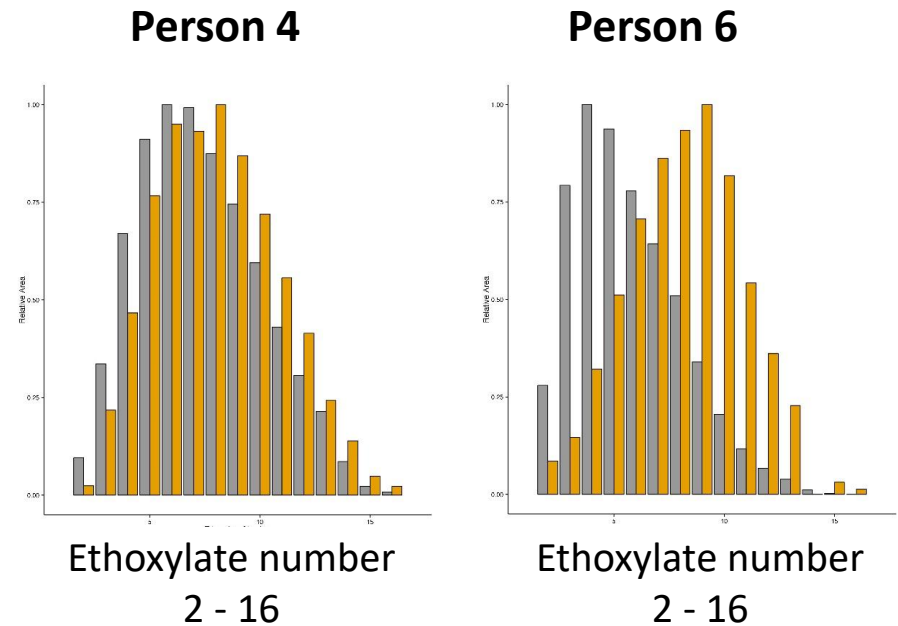


Nonionic surfactant ethoxymer distributions in paired dust/handwipe samples

Example: NPEO

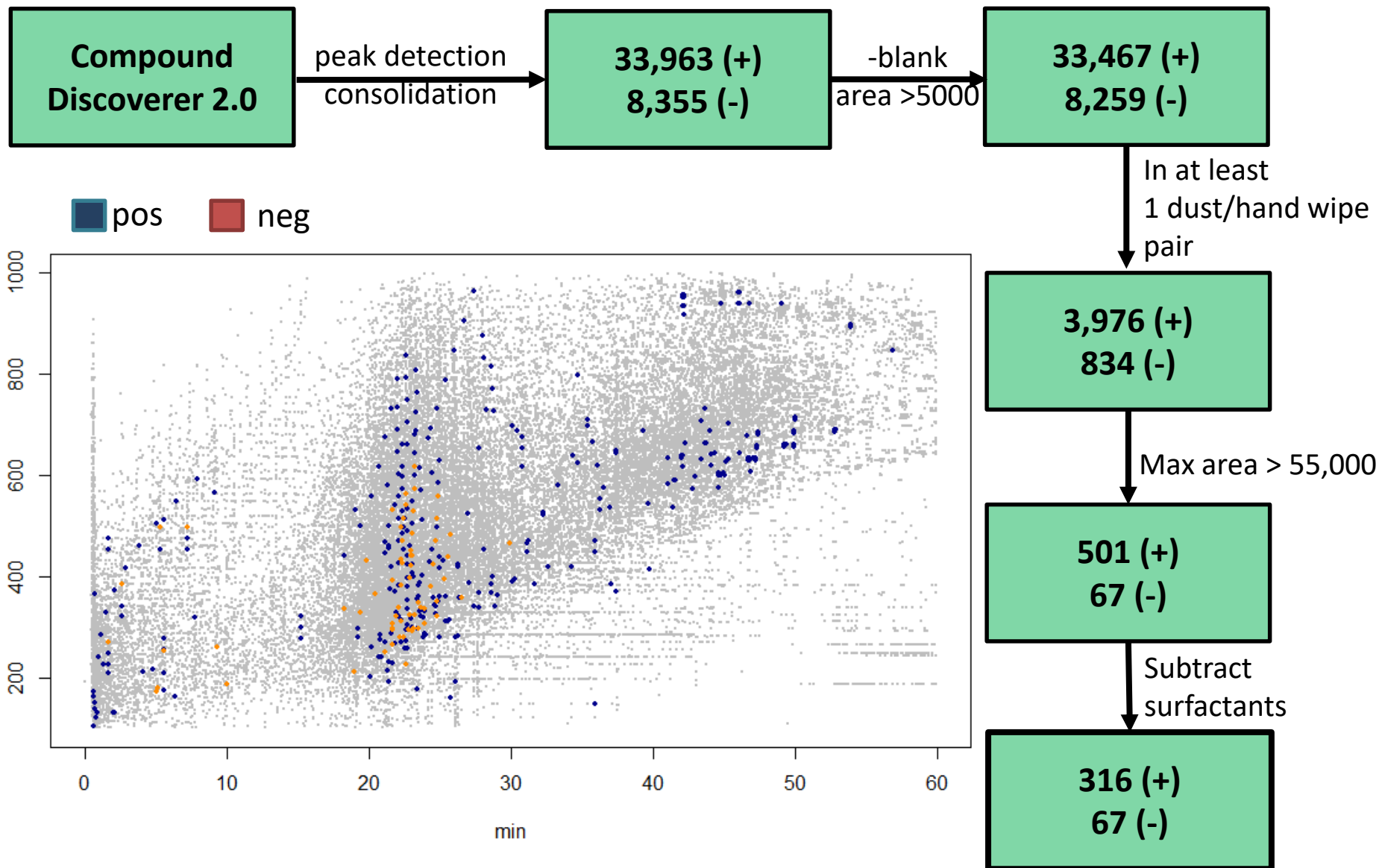


Example: Alcohol Ethoxylate C14

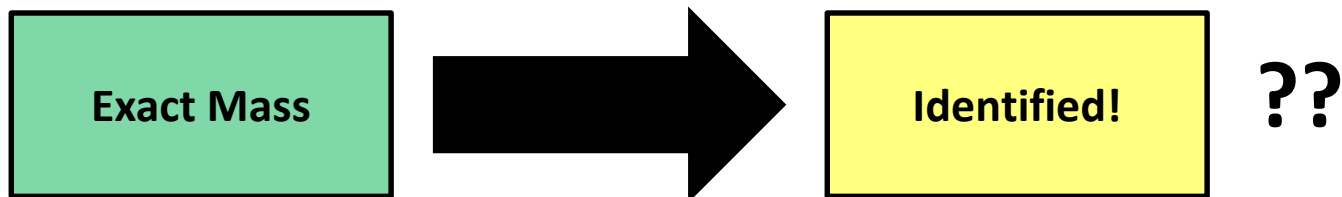


Ethoxymer distribution varied from surfactant to surfactant and person to person – this suggests different sources of ethoxylated surfactants in some cases.

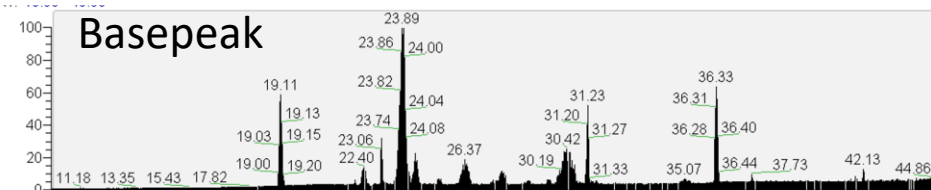
Subtraction of surfactant features prioritizes monomeric compounds for identification



Workflow strategies for identifying compounds in dust/handwipes from LC-HRMS data



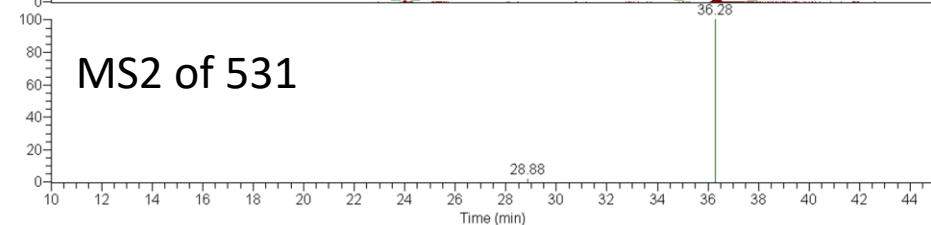
Basepeak



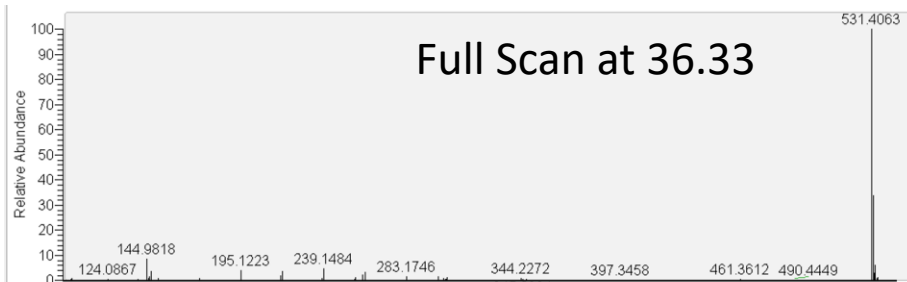
XIC 531.4061



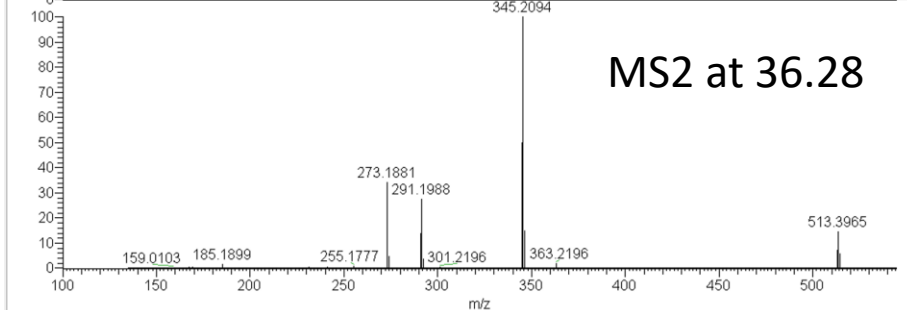
MS2 of 531



Full Scan at 36.33



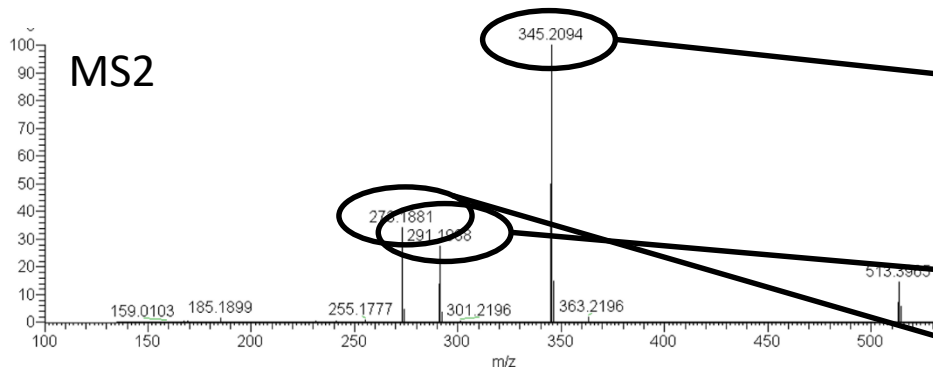
MS2 at 36.28



Molecular formula generation: Vital first step toward structural ID

SIRIUS (<http://bio.informatik.uni-jena.de/software/sirius/>)

Calculates molecular formula assuming that all fragments must be a subset of the parent formula

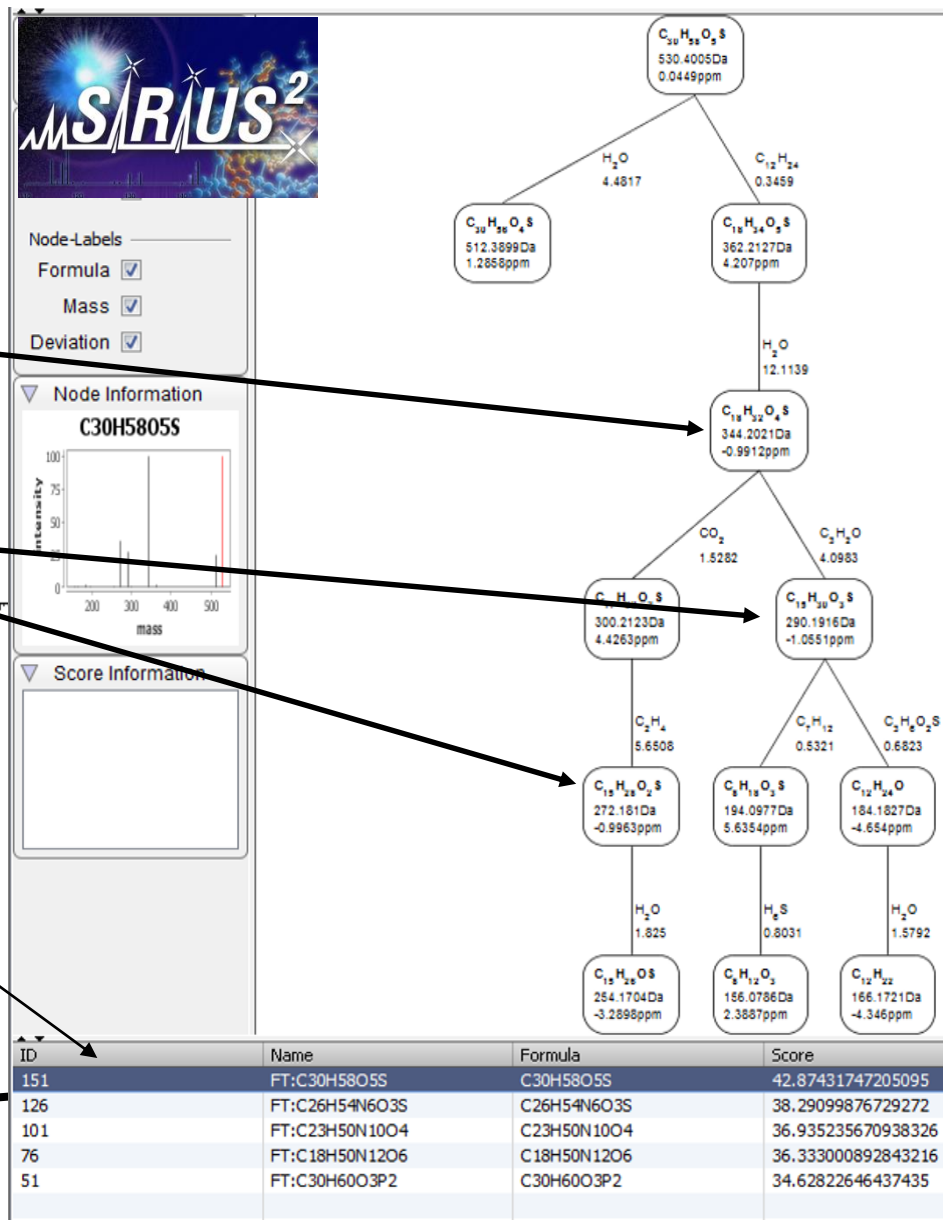


With 5 ppm mass range: **16 possibilities**

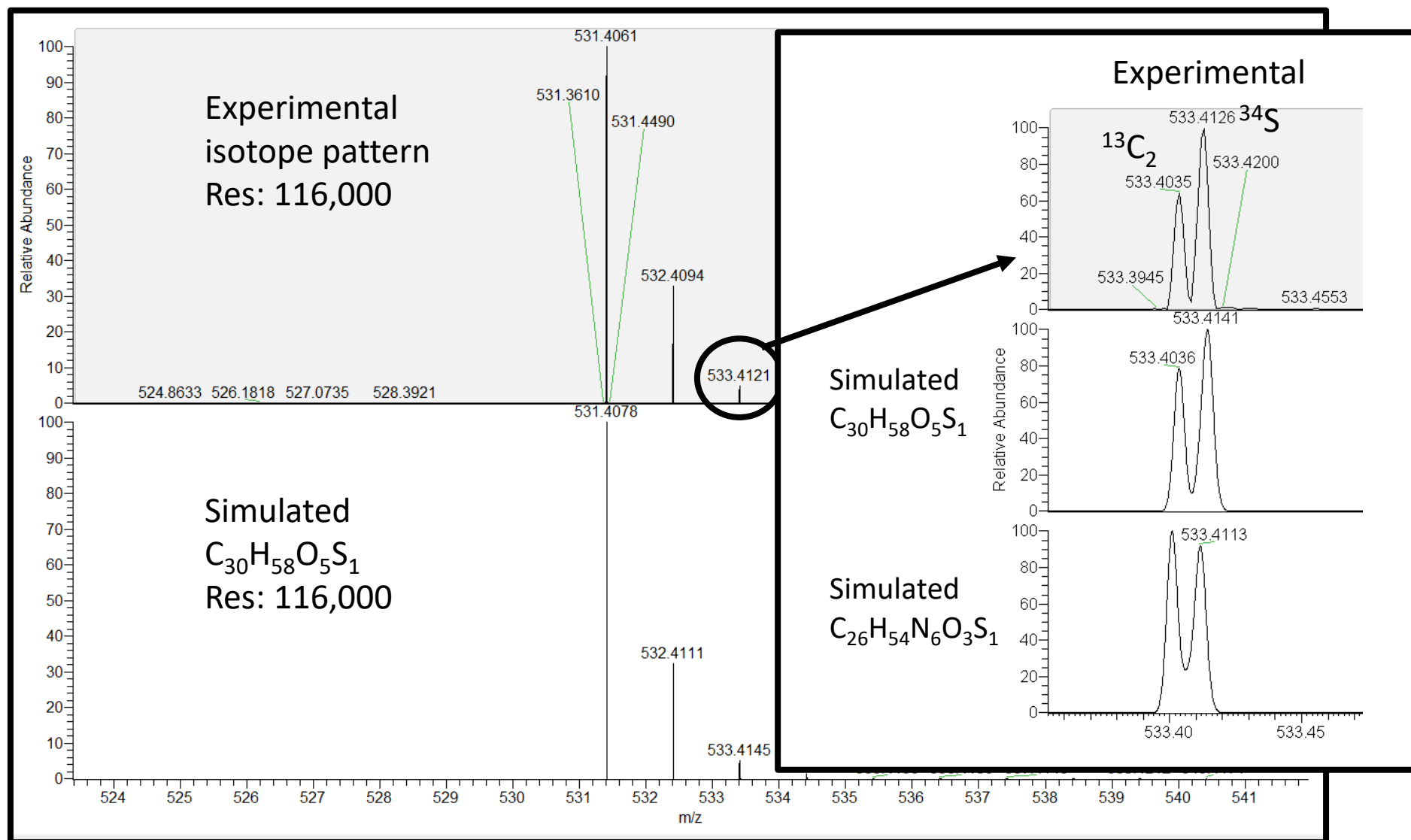
With fragment trees: limited to **5**

Highest scoring molecular formula for m/z

531.4061: **$C_{30}H_{58}O_5S_1$**



Ultra-high resolution allows molecular formula validation by isotope fine structure inspection



Molecular Formula $C_{30}H_{58}O_5S$ SciFinder database search

Explore ▾ Saved Searches ▾ SciPlanner

Molecular Formula "C30 H58 O5 S" > substances (6)

Save Print Export

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0 of 6 Substances Selected

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Substance Role ▾
Preparation 4
Reactant or Reagent 4
Properties 2
Analytical Study 1
Formation, Nonpreparative 1
Process 1
Uses 1

Show More

1. **17243-14-0** [?]
~13 ~3

 $C_{30}H_{58}O_5S$
Propanoic acid, 3,3'-sulfinyldi-, 1,1'-didecyl ester
[Regulatory Information](#)

2. **26826-06-2** [?]
~2

 $C_{30}H_{58}O_5S$
Valeric acid, 5,5'-sulfinyldi-, didecyl ester (8CI)
[Experimental Properties](#)

3. **131986-20-4** [?]
~1

 $C_{30}H_{58}O_5S$
2-Propanol, 1-[(12-cyclohexyldodecyl)oxy]-3-[[2-[2-[(tetrahydro-2H-pyran-2-yl)oxy]ethoxy]ethyl]thio]-

4. **158977-98-1** [?]
~1

 $C_{30}H_{58}O_5S$
[Detailed description of the molecule's stereochemistry and functional groups]
[Regulatory Information](#)

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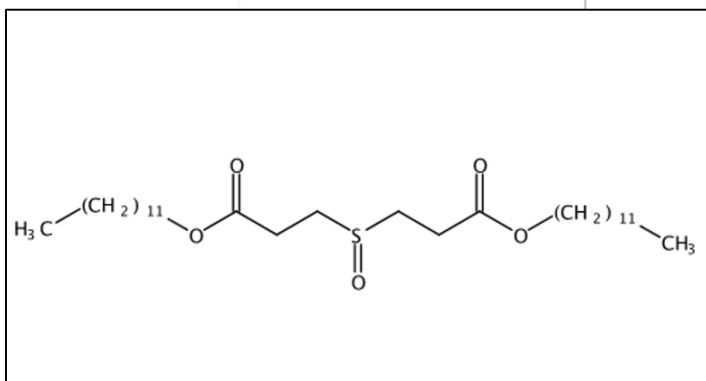
- ☒ Adverse Effect, including toxicity
- ☒ Analytical Study
- ☐ Biological Study
- ☐ Combinatorial Study
- ☐ Crystal Structure
- ☐ Formation, nonpreparative
- ☐ Miscellaneous
- ☒ Occurrence
- ☐ Preparation
- ☐ Process
- ☐ Properties
- ☐ Prophetic in Patents
- ☐ Reactant or Reagent
- ☐ Spectral Properties
- ☐ Uses

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Molecular Formula "C30 H58 O5 S" > substances (6) > 17243-14-0 > **get references (2)**
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Bakhtiar Ray 1

Doss George A 1

Foniokova E 1

Franklin Ronald B 1

Pac J 1

Patel Shefali 1

Sedlar J 1

Xia Yuan Qing 1

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☐ 1. **Identification of a new source of interference leached from polypropylene tubes in mass-selective analysis**
[Quick View](#) [Other Sources](#)

By Xia, Yuan-Qing; Patel, Shefali; Bakhtiar, Ray; Franklin, Ronald B.; Doss, George A.

From Journal of the American Society for Mass Spectrometry (2005), 16(3), 417-421. | Language: English, Database: CAPLUS

An interference leached from polypropylene tubes was identified to be a sulfoxide oxidative product of didodecyl 3,3'-thiodipropionate (DDTDP) that was used to prevent oxidative degrdn. of synthetic polymers. A sulfone oxidative product of DDTDP leached from the polypropylene tubes was also obsd. The interfering compds. were isolated by LC and characterized using time-of-flight mass spectrometry and NMR. Authentic sulfoxide and sulfone products of DDTDP were also prepd. by reacting DDTDP with hydrogen peroxide reaching an unequivocal structural assignment. In conclusion, when analytes of i...

☐ 2. **Gas-chromatographic determination of dilauryl β,β' -thiodipropionate and its primary oxidation products**
[Quick View](#) [Other Sources](#)

By Sedlar, J.; Foniokova, E.; Pac, J.

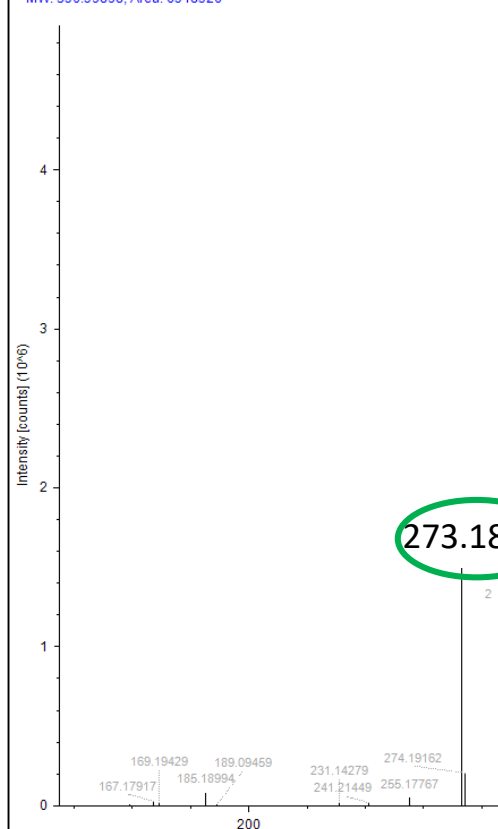
From Analyst (Cambridge, United Kingdom) (1974), 99(1174), 50-3. | Language: English, Database: CAPLUS

The title compd., and dilauryl sulfonyl- β,β' -dipropionate and sulfonyl- β,β' -dipropionate, were detd. by hydrolysis in 5N MeOH-KOH and gas chromatog. detn. of the lauryl alc. extd. by CHCl_3 using n-octadecane as internal marker on a 6 ft glass 1.5% fluorosilicone oil FS-1265 or Chromosorb W AW-DMCS column at 165°. The detn. gave results consistently low by a factor of 0.97. Detns. were of ≥ 10 - $\mu\text{g/ml}$ concns. with a std. deviation of $\pm 3\%$.

Identification of a New Source of Interference Leached from Polypropylene Tubes in Mass-Selective Analysis

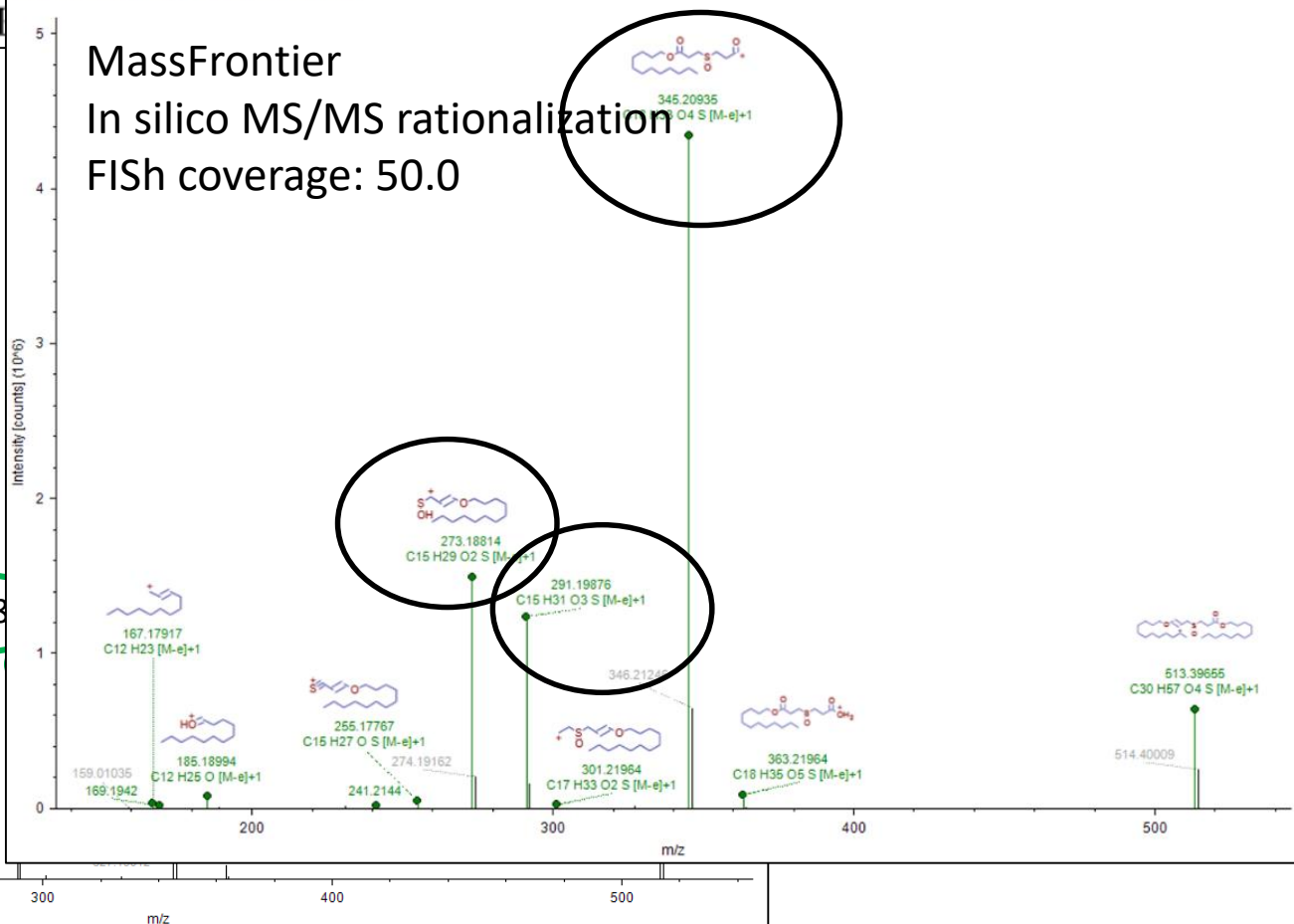
Yuan-Qing Xia, Shefali I

#3385, RT=36.227 min, FTMS (+), MS2 (CID, DDF, m/z=531.41, z=+1)
MW: 530.39898, Area: 6318326

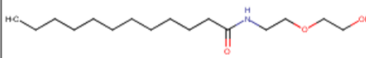
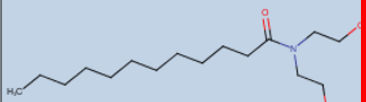
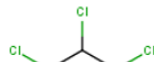
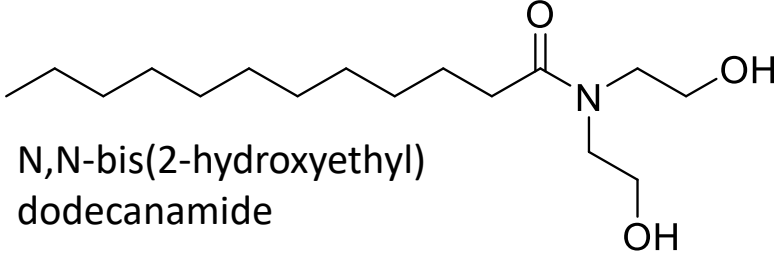
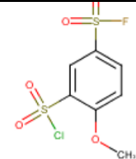
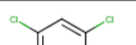


#3385, RT=36.227 min, FTMS (+), MS2 (CID, DDF, m/z=531.41, z=+1)
n/a C30 H58 O5 S, MW: 530.40050, Area: 6318326
FISH Coverage: 11 Direct, 5 Unmatched, 4 Skipped

MassFrontier
In silico MS/MS rationalization
FISH coverage: 50.0



Identifying features from an in-house curated suspect database (31,985 entries)

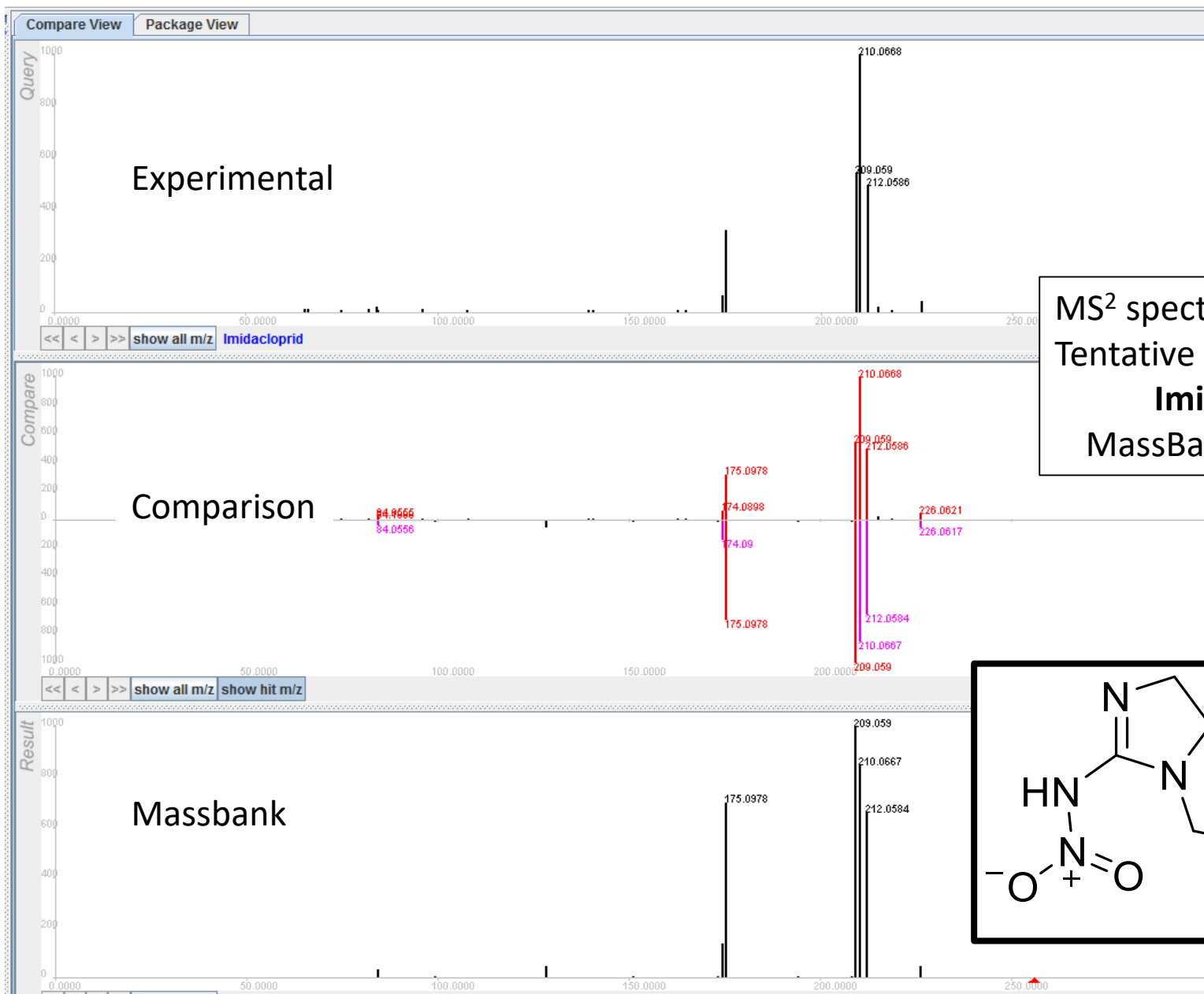
Duke University Unique Record #	Structure	Formula	Exact Mass [▲] (1)	CAS #	TSCA #	Pesticide Product Code #	Traditional Name	CAS Index Name	LogP	2012 National Production Volume	EFS #
DU15264		C16H33NO3	287.24603	20138-28-7	TSCA21318	PC79067	N-[2-(2-hydroxyethoxy)ethyl]dodecanamide	Dodecanamide, N-[2-(2-hydroxyethoxy)ethyl]-	3.16		
DU3741		C16H33NO3	287.24603	130-40-1	TSCA2977	PC79018	N,N-bis(2-hydroxyethyl)dodecanamide	Dodecanamide, N,N-bis(2-hydroxyethyl)-		1,000,000 - 10,000,000	
DU177		C6H6Cl6	287.86008	55-98-9	TSCA182	PC9001	lindane	Cyclohexane, 1,2,3,4,5,6-hexachloro-, (1.alpha.,2.alpha.,3.beta.,4.alpha.,5.alpha.,6.beta.)-	4.35		
DU18											
DU27			287.93292				methoxybenzenesulfonyl fluoride	chlorosulfonyl-4-methoxy-	1.42		
DU10120		C12H7Cl3O2		3380-34-5	TSCA11317	PC54901	tricosan	Phenol, 5-chloro-2-(2,4-dichlorophenoxy)-			EFS9

Search by formula

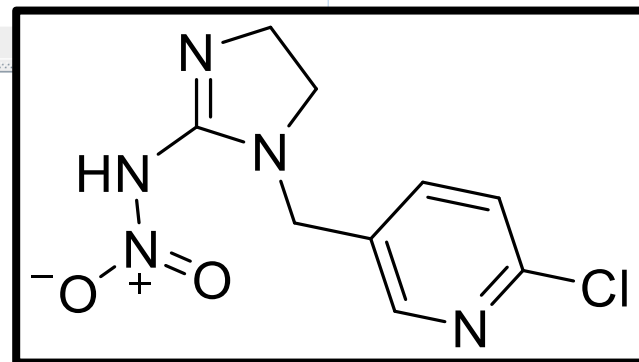
2012 National Production Volume

Tentative identification supported by in silico MS² prediction using Mass Frontier (FISh Score: 80)

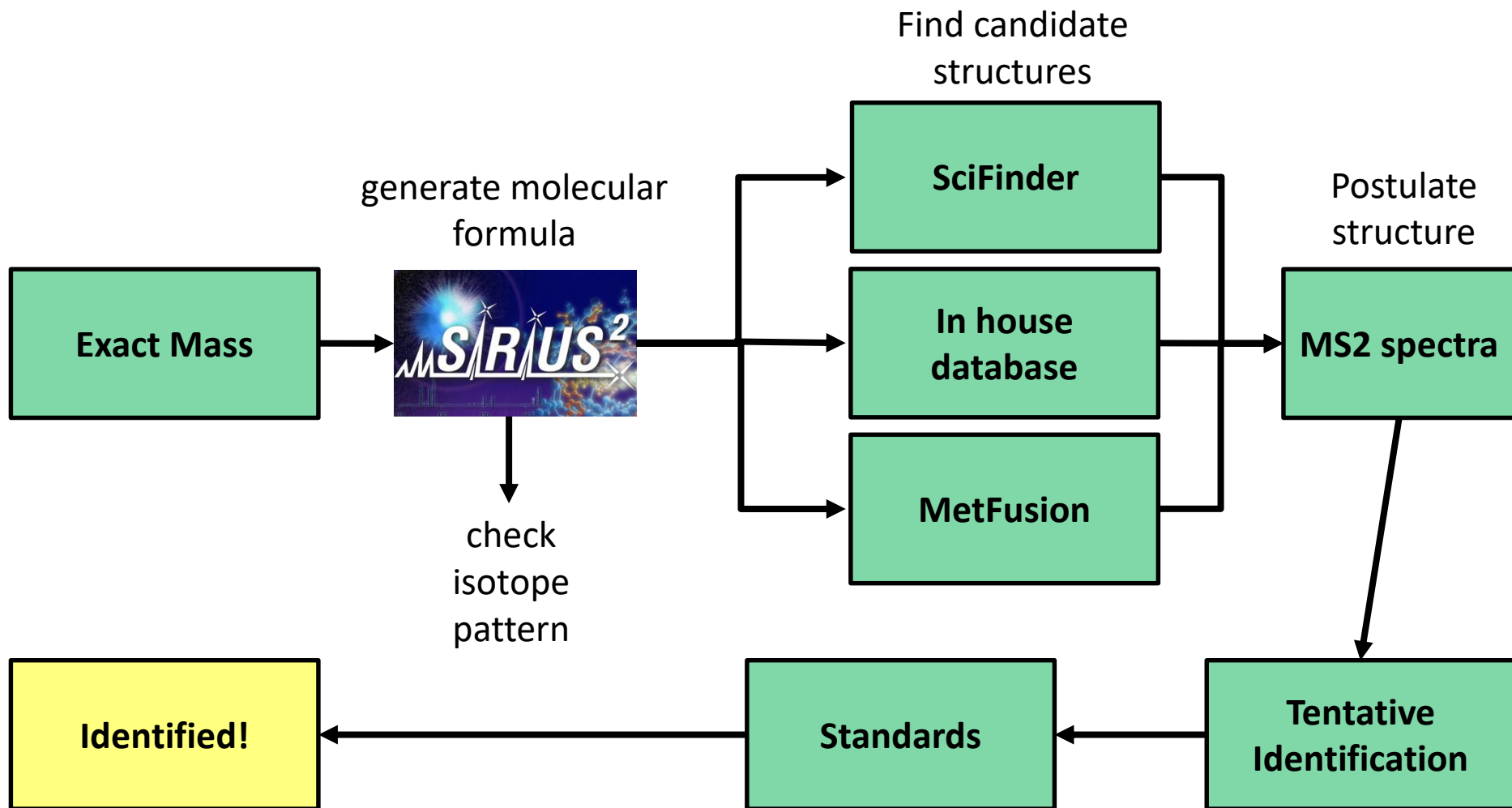
MetFusion for compound ID from HRMS² data



MS² spectrum is a match
Tentative identification:
Imidacloprid
MassBank Score: 0.96



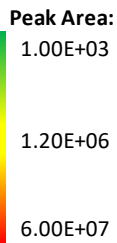
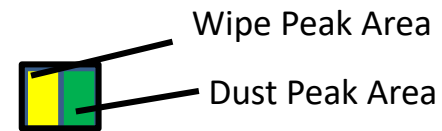
Generalized workflow strategies for identifying SVOC contaminants in paired dust/handwipes by LC-HRMS



Compounds identified in dust/handwipes

34 compounds

- 10 identified with Standard
- 24 tentatively identified



Name		# of hits		Paired samples										STD?
		W	D	1	2	3	4	5	6	7	8	9	10	
Organophosphates	di-tertbutyl triphenyl phosphate	4	5											
	tri-(2-butoxyethyl)-phosphate (TBOEP)	10	10											x
	tris (4-butyl-phenyl) phosphate (TBPP)	7	5											x
	tris (2-chloro-ethyl) phosphate (TCEP)	5	10											x
	tris (1-chloro-isopropyl) phosphate (TCPP)	10	10											x
	tricresyl phosphate	6	7											
	triphenyl phosphate (TPP)	10	10											x
	V6	2	4											x
Surfactants	dodecyl sulfate	2	5											
	tridecyl sulfate	2	6											
	tetradecyl sulfate	2	5											
	pentadecyl sulfate	3	6											
	hexadecyl sulfate	6	10											
	dodecylethanolamine	9	8											
	N-lauroyl sarcosine	6	6											
	perfluorooctanesulfonic acid (PFOS)	0	2											

Surfactants used in
shampoo, cosmetics

Peak Area:

1.00E+03

1.20E+06

6.00E+07

Peak Area:

1.00E+03

1.20E+06

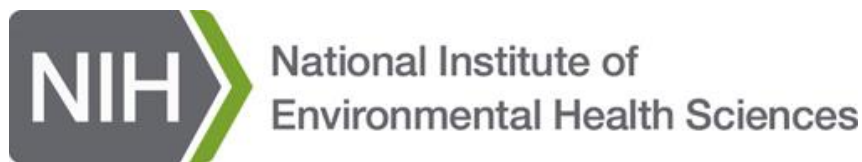
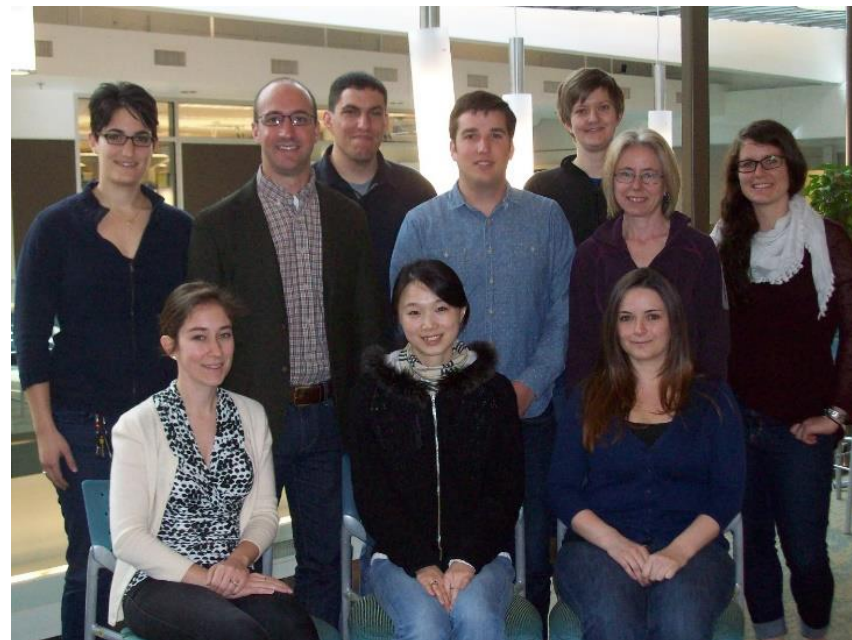
6.00E+07

Conclusions: Exploring the indoor environment exposome using non-targeted analysis strategies

- (2D)LC-HRAM mass spectrometry is a powerful tool for analysis of SVOC compounds in dust and hand wipe samples.
- Non-targeted workflows allow a more holistic view of contaminant exposure in indoor environments relative to targeted analysis.
- 213 tentative and confirmed identifications were made from 567 filtered components in dust/wipes (37.5% of filtered features).
- The most dominant compounds in dust and handwipes were non ionic surfactants such as nonylphenol ethoxylates or alcohol ethoxylates.

Acknowledgement

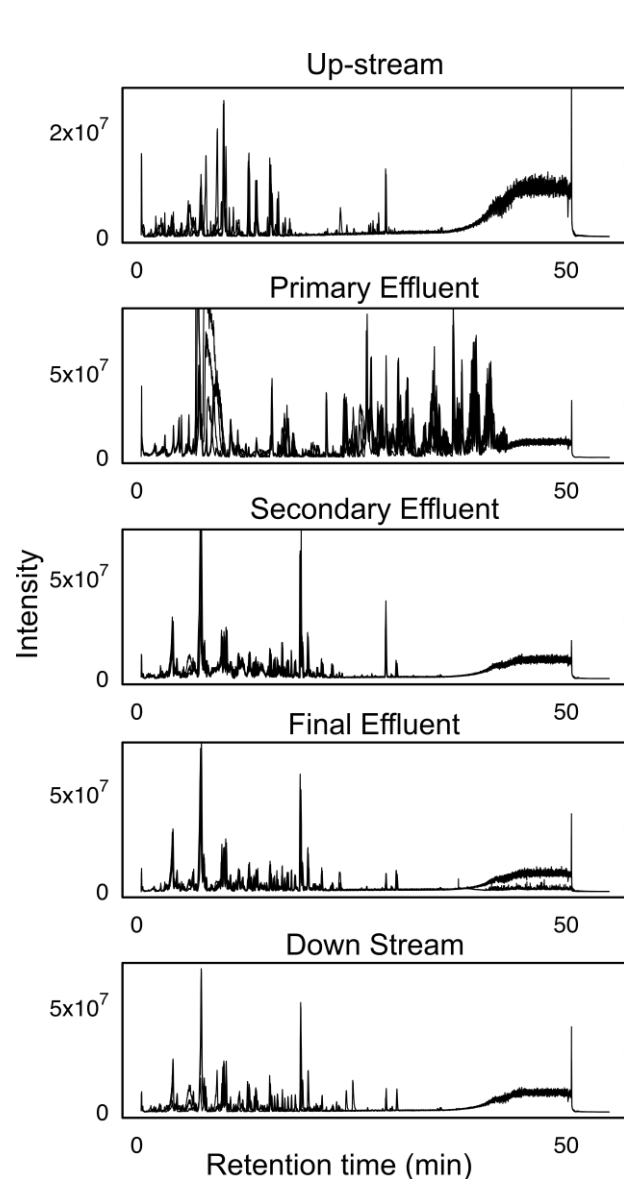
Ferguson Lab Group
Stapleton Lab Group



ThermoFisher
SCIENTIFIC

Richard Jack and Dipankar Ghosh

Data analysis workflow



Compound Discover 2.0 (Thermo)

- Retention time alignment
- Peak detection
- De-isotope and de-adduct
- Feature consolidation
- Gap filling

Consolidated Peak Table

Mol. wt.	RT	Measured Area			
		Samp. 1	Samp. 2	Samp. 3	Samp n
x_1	y_1	$z_{1,1}$	$z_{1,2}$	$z_{1,3}$	$z_{i,1}$
\vdots	\vdots	\vdots	\vdots	\vdots	\vdots
x_i	y_i	$z_{i,1}$	$z_{i,2}$	$z_{i,3}$	$z_{i,n}$